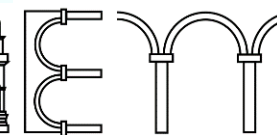
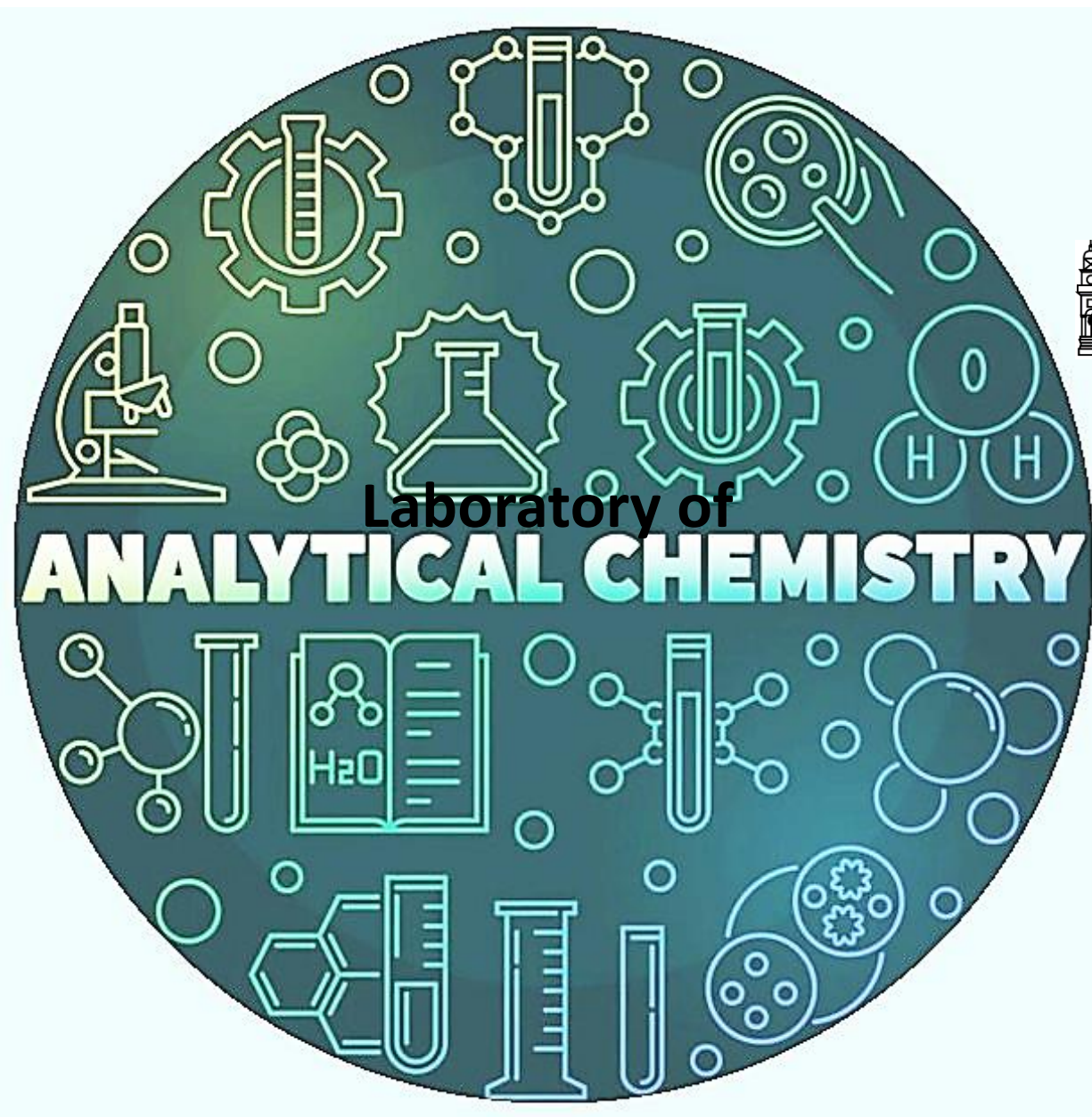




**Department of Chemistry**



ΙΝΣΤΙΤΟΥΤΟ  
ΕΠΙΣΤΗΜΗΣ ΥΛΙΚΩΝ  
ΚΑΙ ΥΠΟΛΟΓΙΣΜΩΝ

**Dimosthenis L. Giokas (Assoc. Prof.)**

# Equipment and Instrumentation

**SHIMADZU**  
FAAS / ETAAS / HGAAS



**SHIMADZU**  
HPLC UV/FL/CONDUCTIVITY



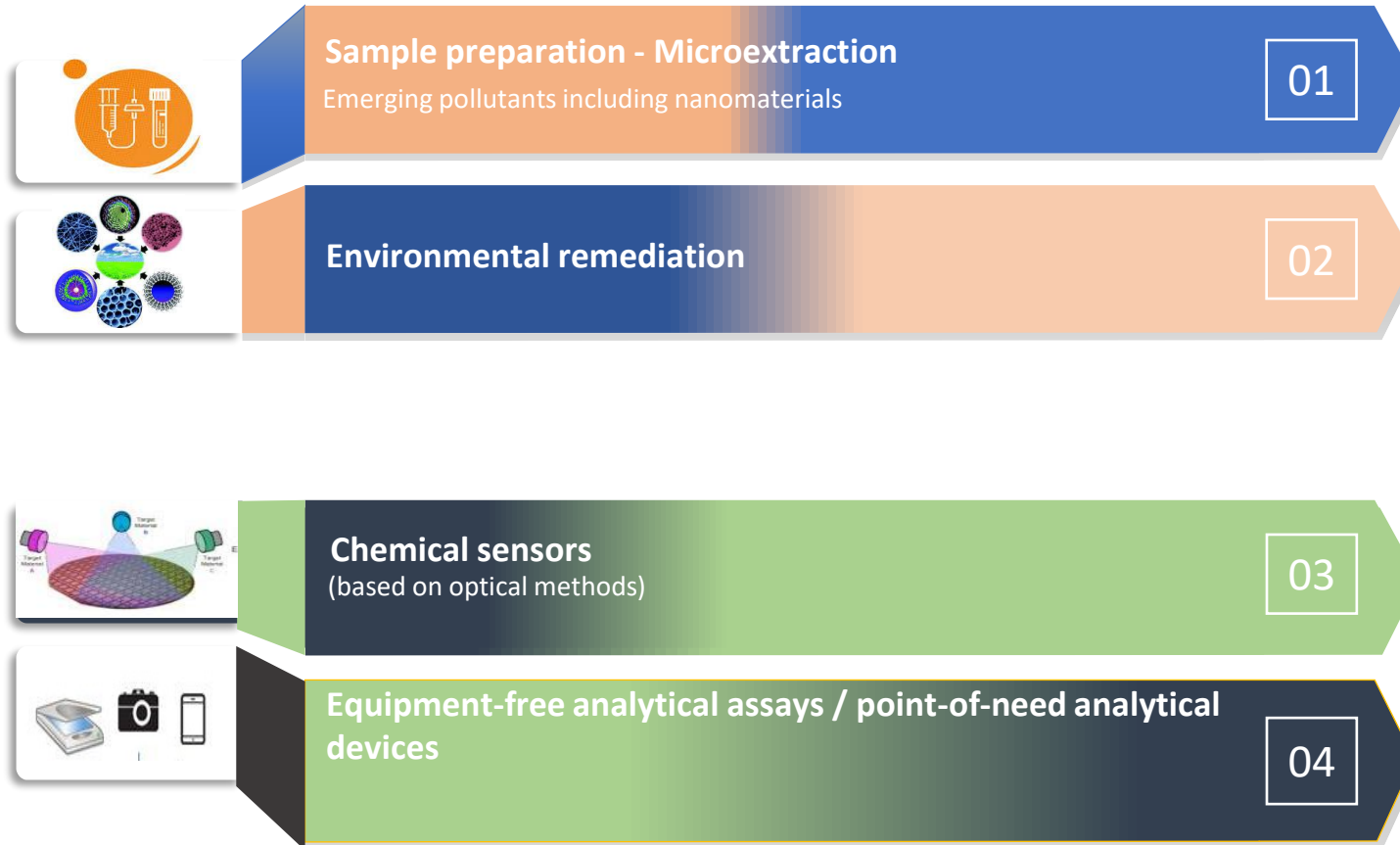
**JENWAY**  
UV-VIS

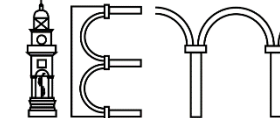


# Equipment and Instrumentation





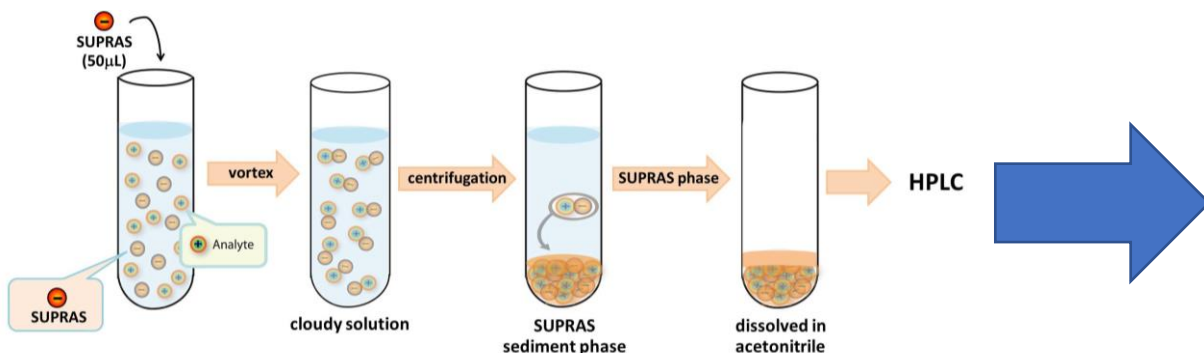




Principle: Adsorption and desorption of analytes

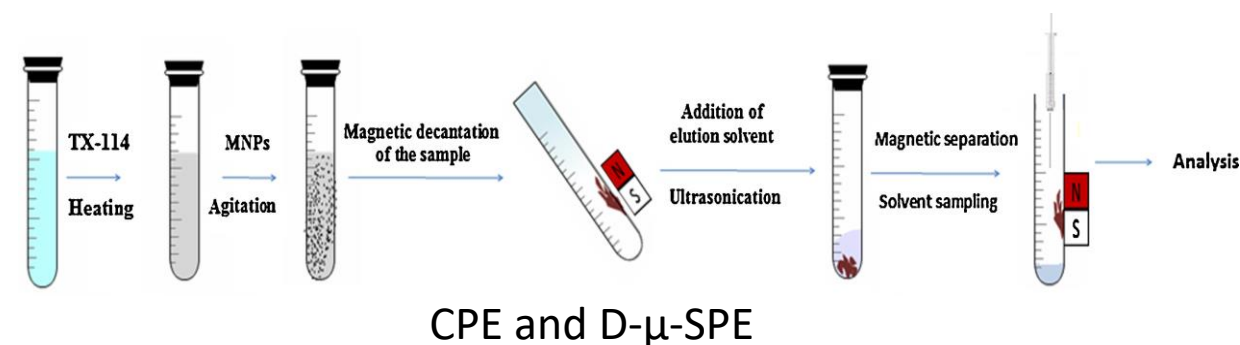
### Common approach

Use of micro/nano sorbents for extraction / microextraction



### Our approach

Combined microextraction methods



Journal of Chromatography A 1251, 2012, 33-39

Microchimica Acta 180, 2013, 775-782

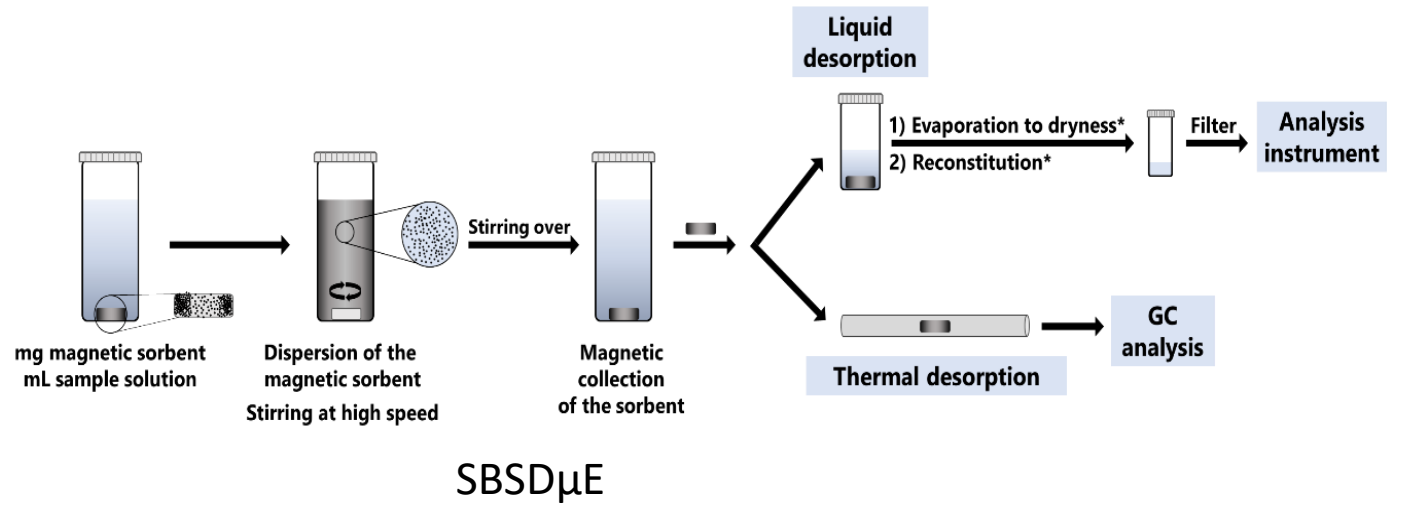
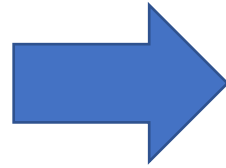




SBSE



D-μSPE



Journal of Chromatography A 1362, 2014, 25-33

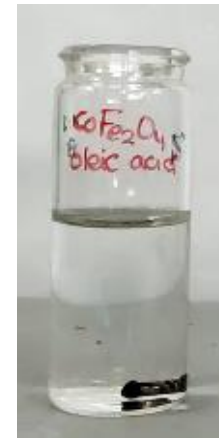
Highlighted Article in Microextraction Tech

<http://microextraction.blogspot.com/2014/11/stir-bar-sorptive-dispersive.html>

Talanta 147, 2016 246-252

Analytica Chimica Acta, 926, 2016, 63-71

Journal of Chromatography A, 1564, 2018, 25-33

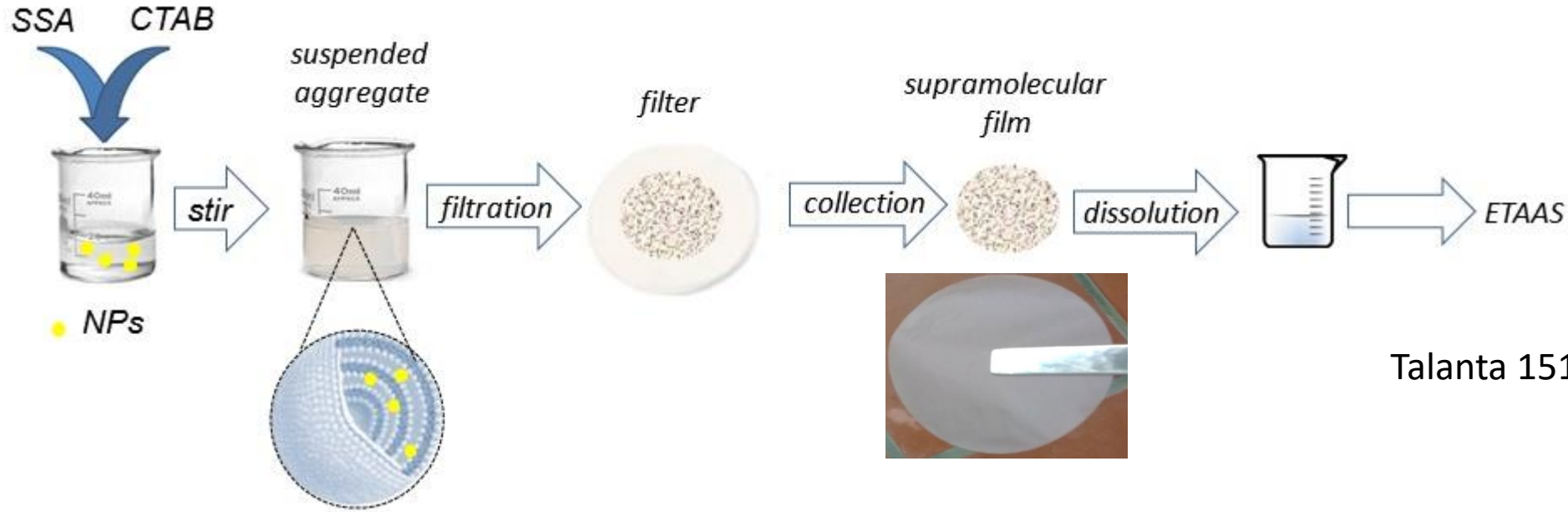


SBSE

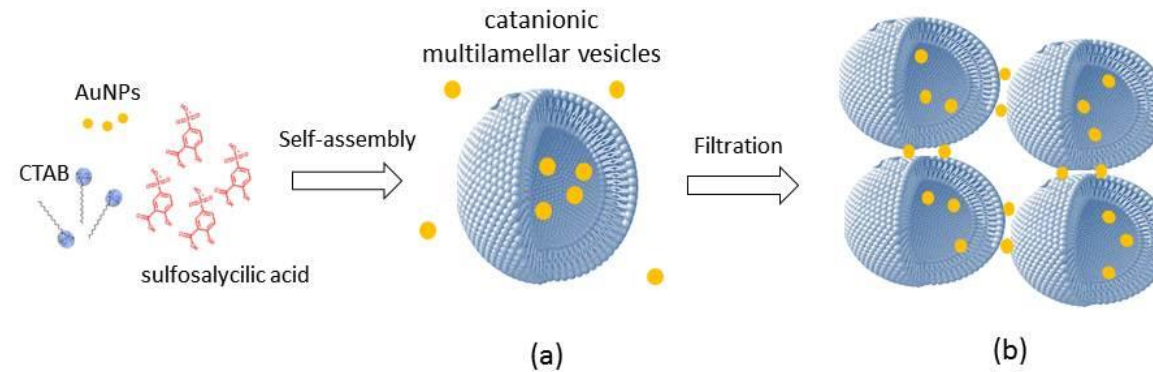
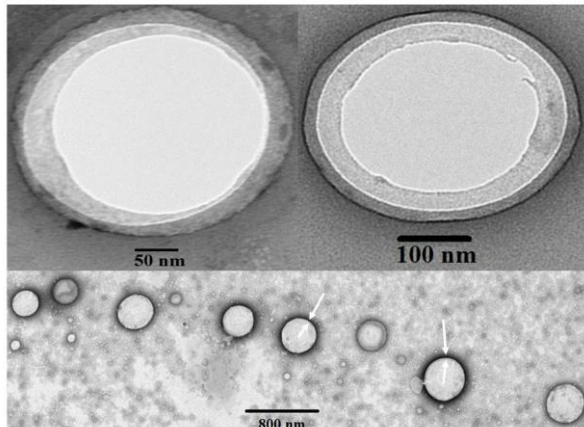


SBDμSPE

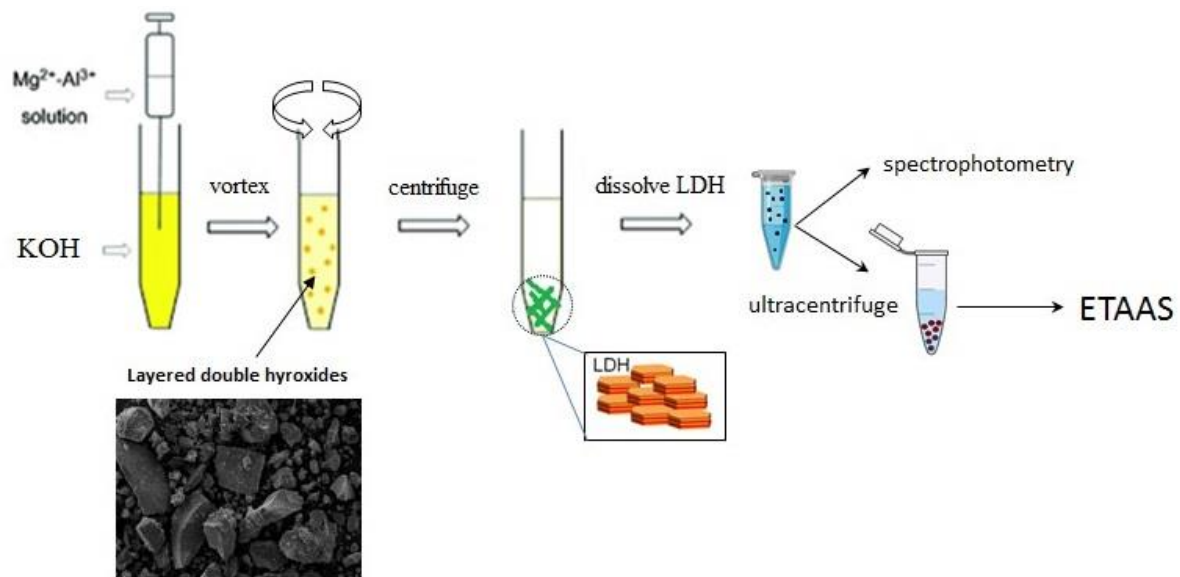
## Microextraction of nanoparticles from environmental samples



Talanta 151, 2016, 91-99

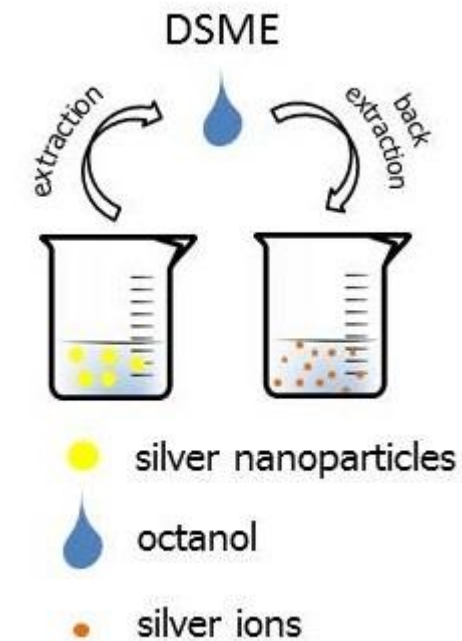


### Microextraction of nanoparticles from environmental samples



Analytical Methods, 12, 2020, 368-375

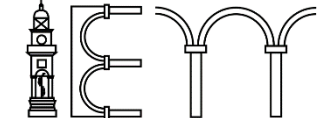
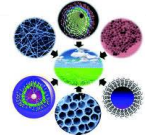
Talanta, 196, 2019, 255-261



Environmental Pollution 263,2020, 114407







## Common approach

Adsorbents (micro/nanomaterials) dispersed in aqueous medium

Batch adsorption studies

Adsorption isotherms in distilled water

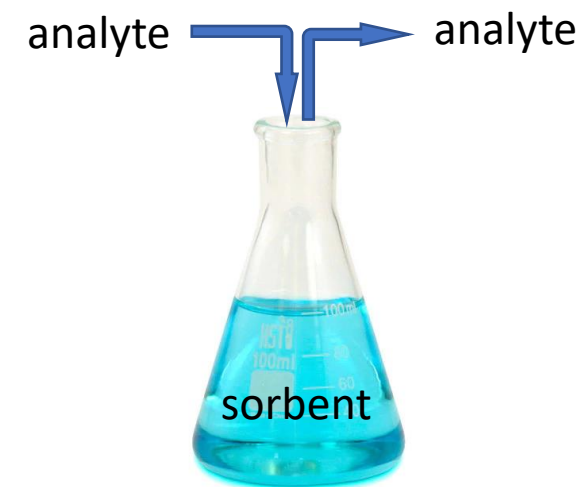


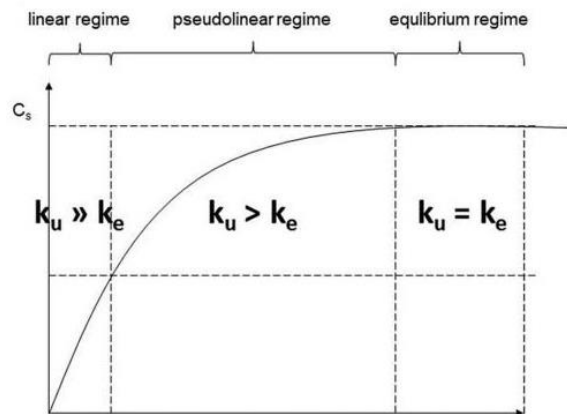
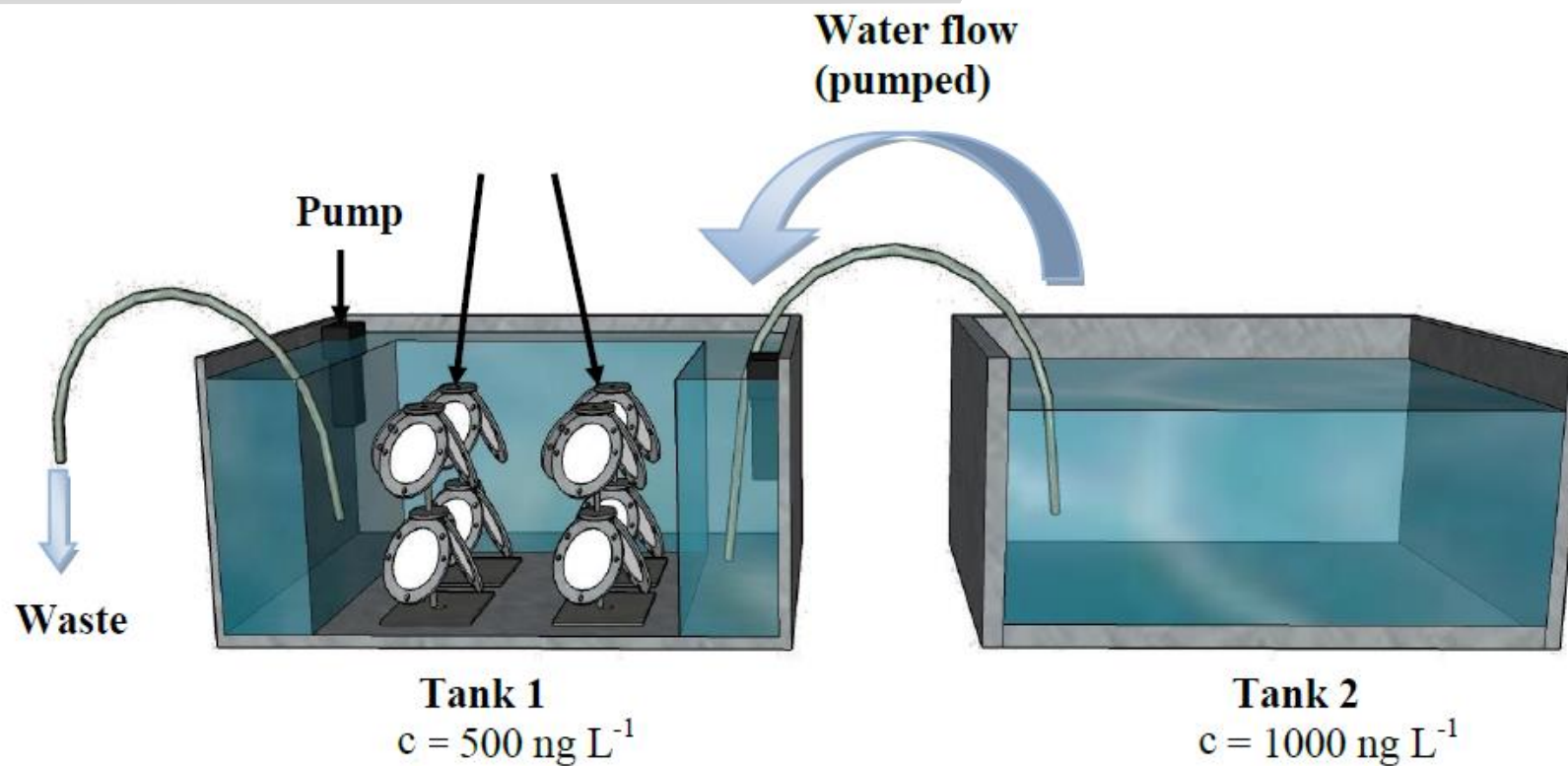
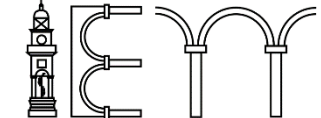
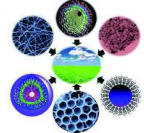
## Our approach

Adsorbents (nanomaterials) immobilized onto solid supports

Real-time adsorption studies

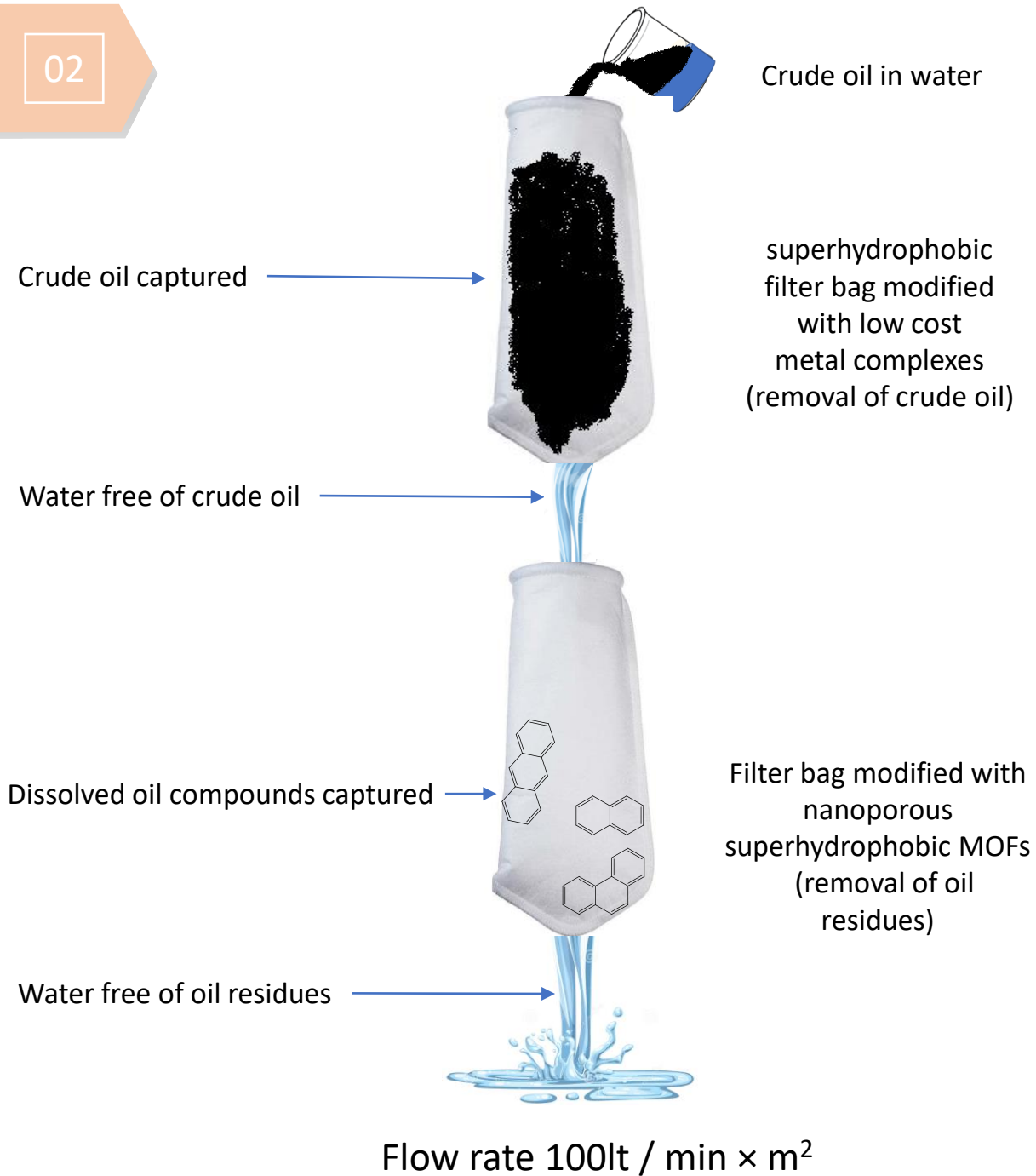
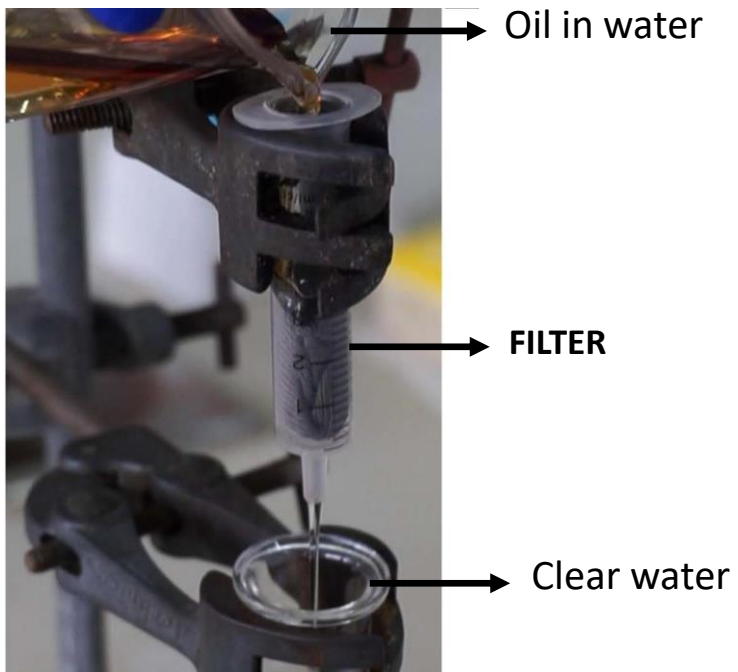
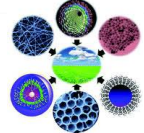
Sampling rate in real samples



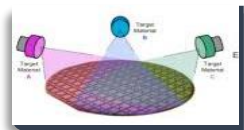


$$R_s = \frac{C_s \times M_s}{C_w \times t}$$

$C_s$  is the analyte concentration in the sorbent (ng/g),  
 $C_w$  is the analyte concentration in the solution (ng/L) after the time  $t$   
 $M_s$  is the mass of sorbent (g).



in cooperation with



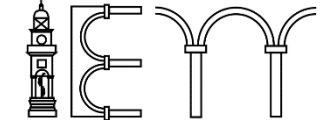
**Chemical sensors**  
(based on optical methods)

03



**Equipment-free analytical assays / point-of-need analytical devices**

04

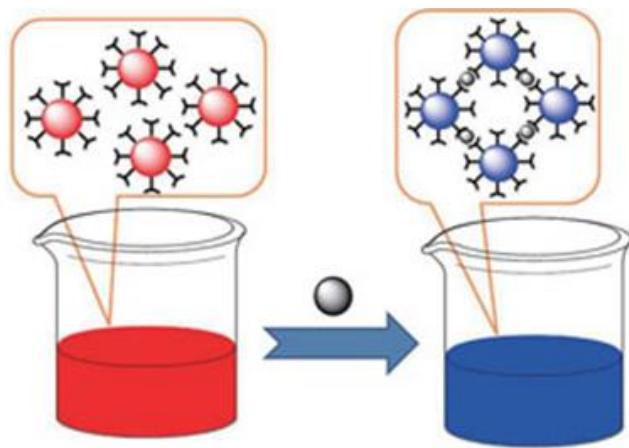


ΙΝΣΤΙΤΟΥΤΟ  
ΕΠΙΣΤΗΜΗΣ ΥΛΙΚΩΝ  
ΚΑΙ ΥΠΟΛΟΓΙΣΜΩΝ

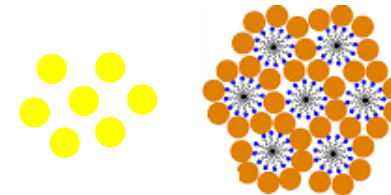
### Our approach

Analyte-mediated formation of nanomaterials

**Common approach**  
Pre-formed nanomaterials



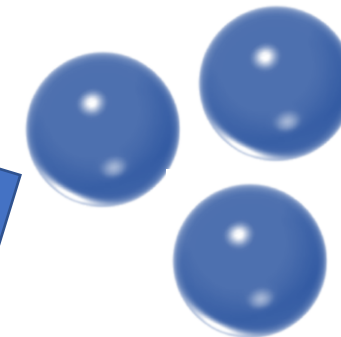
Me<sup>n+</sup> Metal coated micelles



without analyte

Reducing agent  
Metal seeds  
UV-Vis irradiation

analyte



λήψη εικόνας

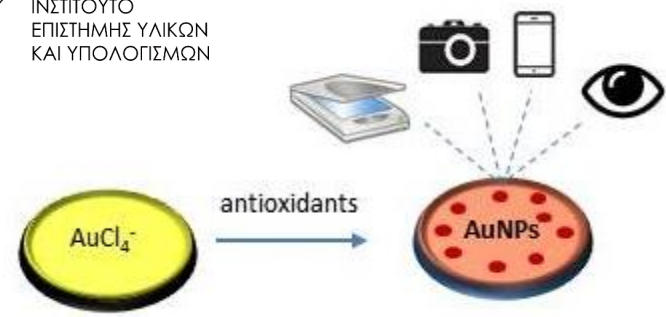
επεξεργασία εικόνας



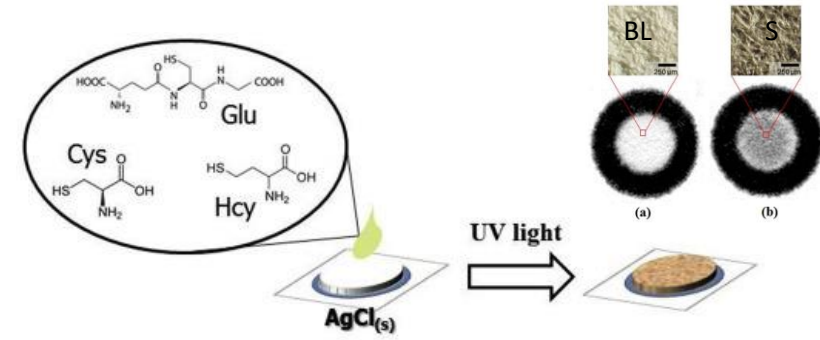


**Chemical sensors**  
 (based on optical methods) 03

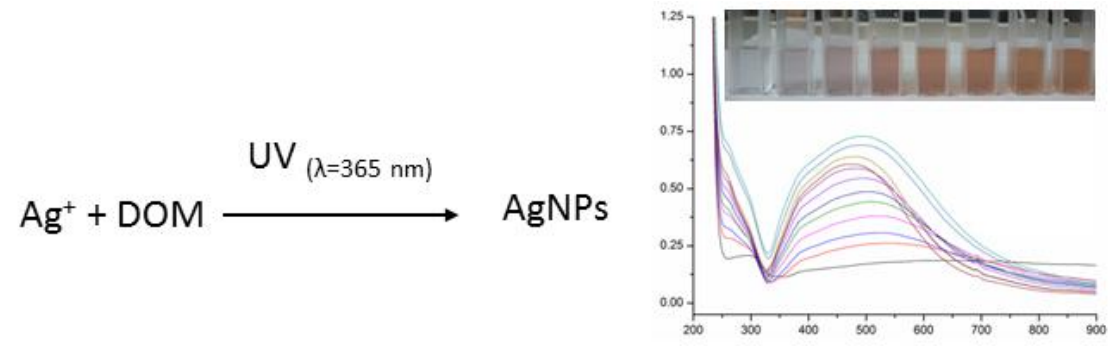
**Equipment-free analytical assays / point-of-need analytical devices** 04



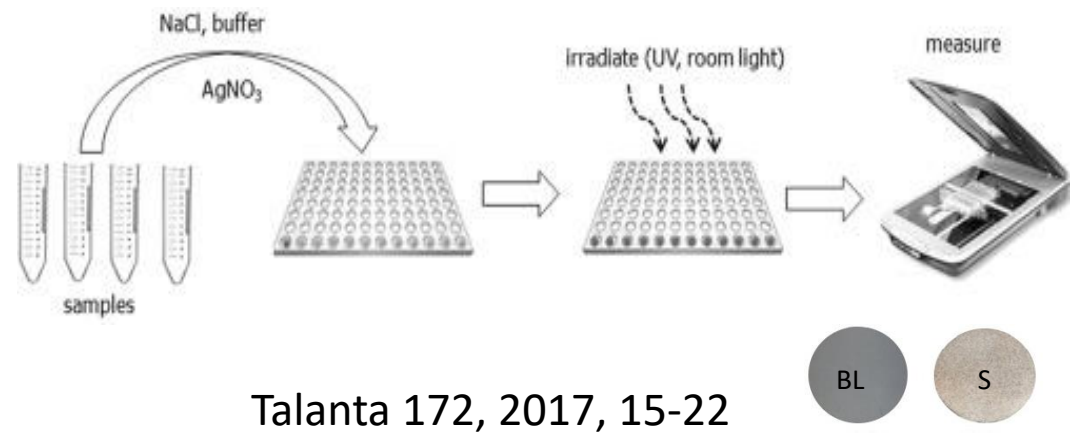
Analytica Chimica Acta 860, 2015, 61-69



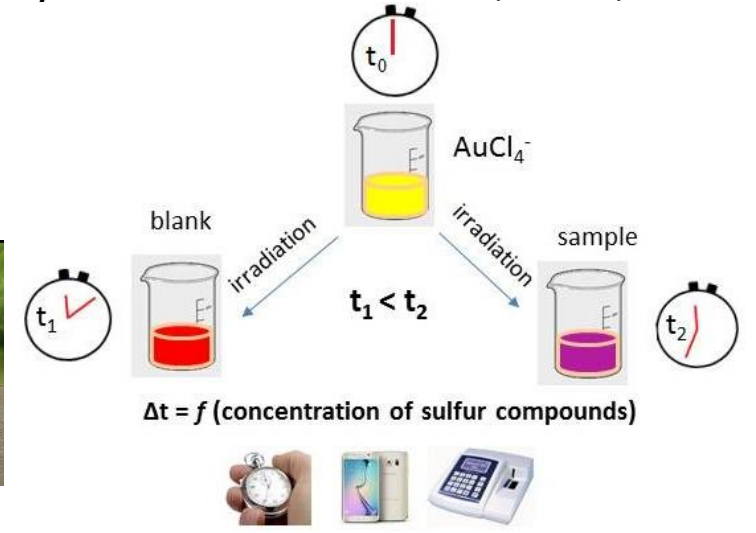
Analytica Chimica Acta 1036, 2018, 89-96



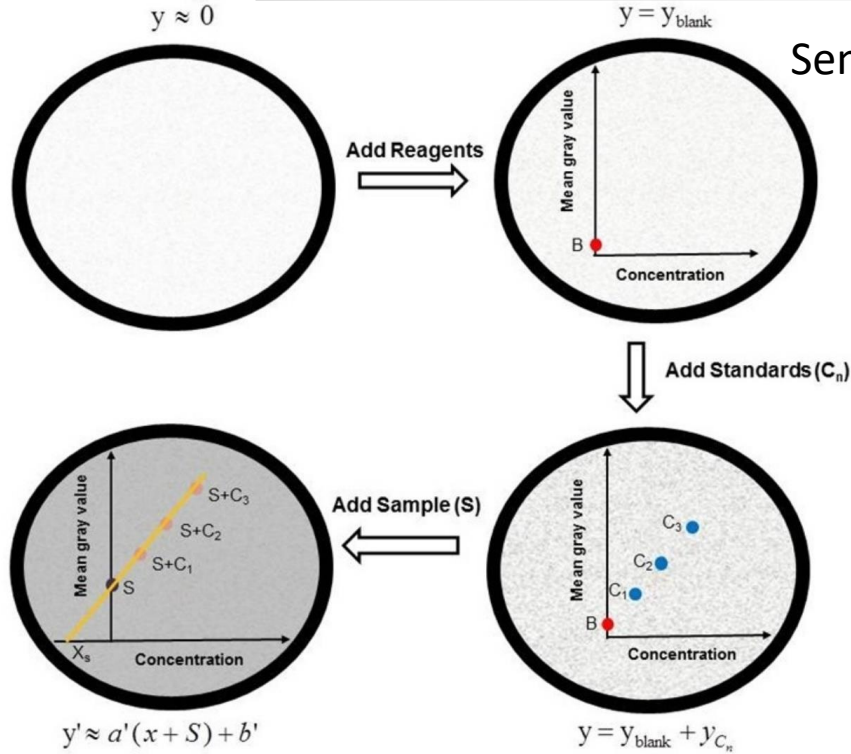
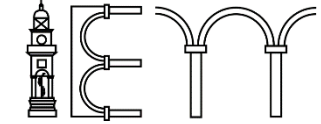
Analytica Chimica Acta 812, 2014, 121-128



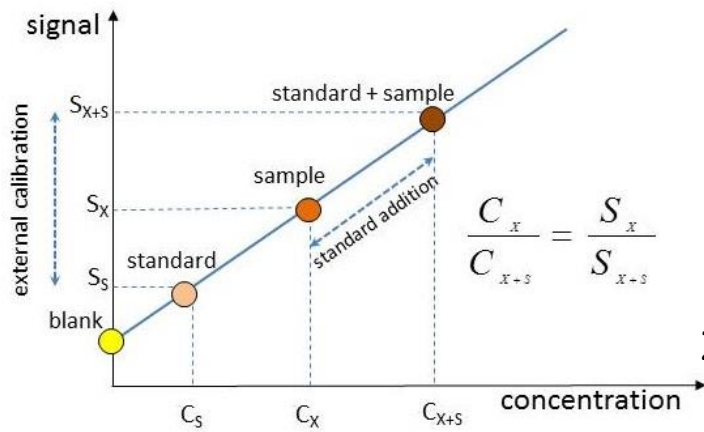
Talanta 172, 2017, 15-22



ACS Omega 2018, 3, 12, 16831–16838

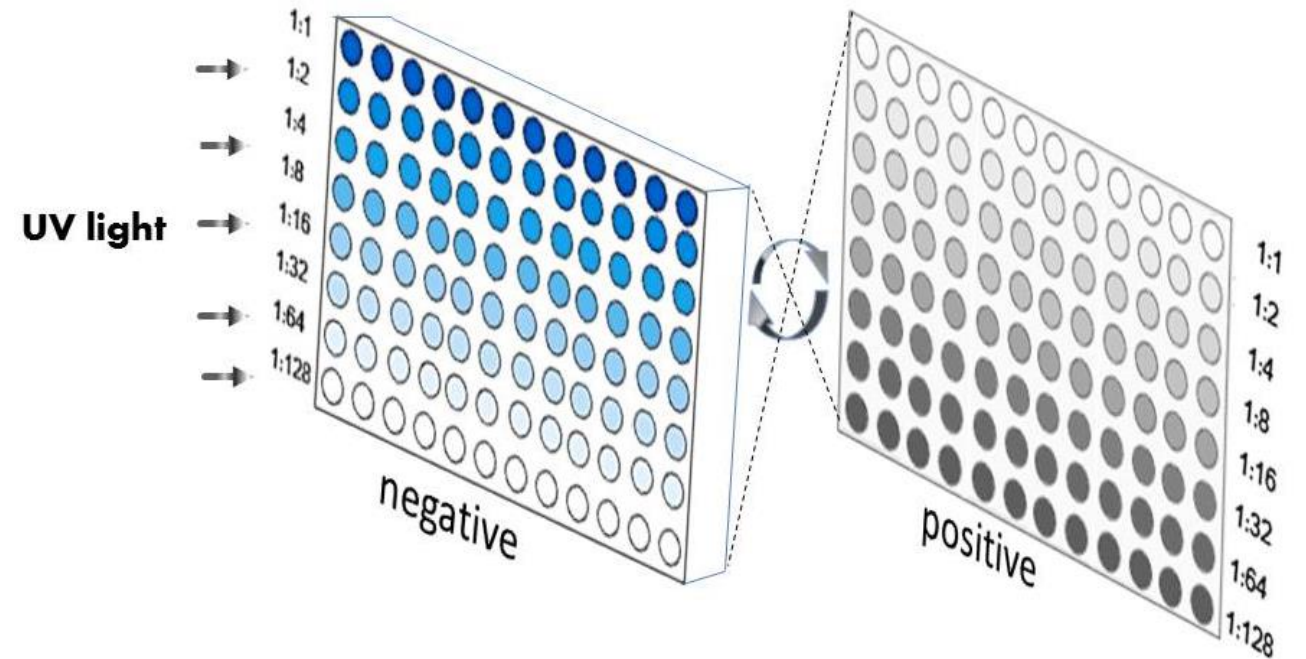


Sensors and Actuators B: Chemical  
253, 2017, 860-867



Talanta  
201, 2019, 149-155

Photography-based photometry



Sensors and Actuators B: Chemical  
328, 2021, 129018

# Our team

- Prof. Athanasios Vlessidis
- Mrs. Tatiana Choleva (PhD)
- Mrs. Charikleia Tziasiou
- Mrs. Vasiliki Gouma
- Mr. Elias Moisiadis
- Mrs. Despoina Gkogkou

