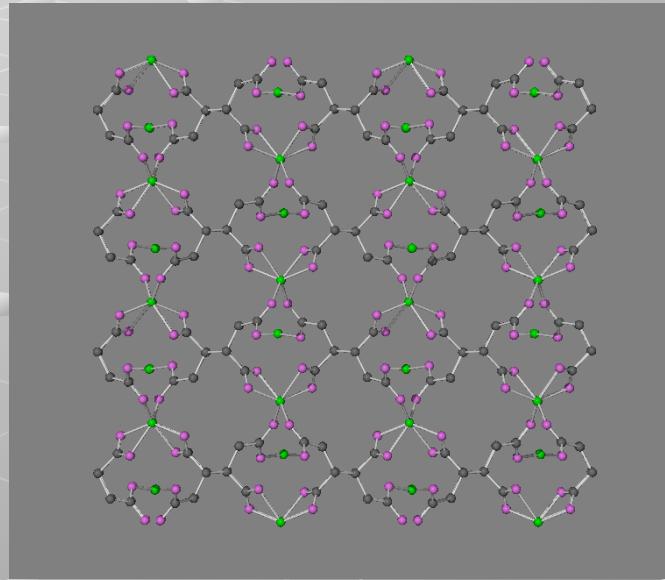




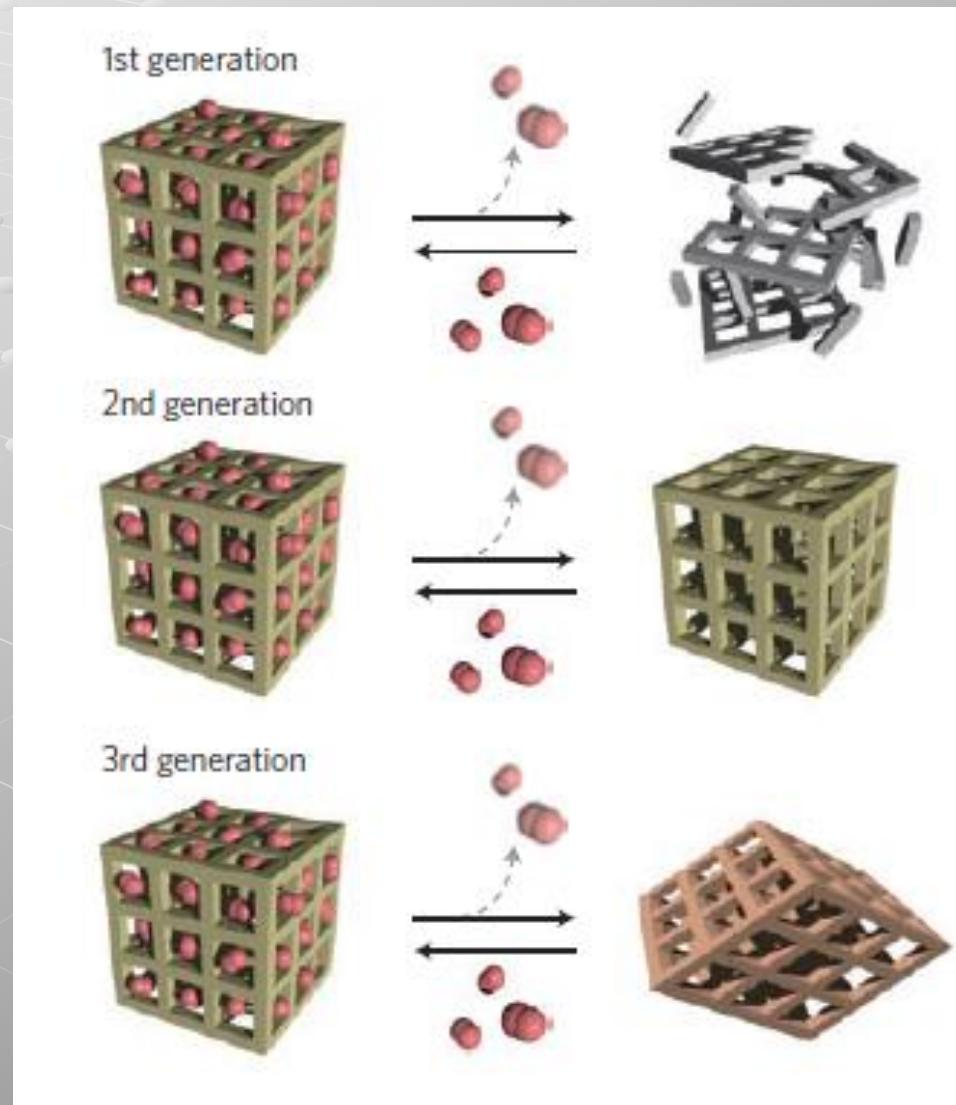
Synthetic Coordination Chemistry



John C. Plakatouras

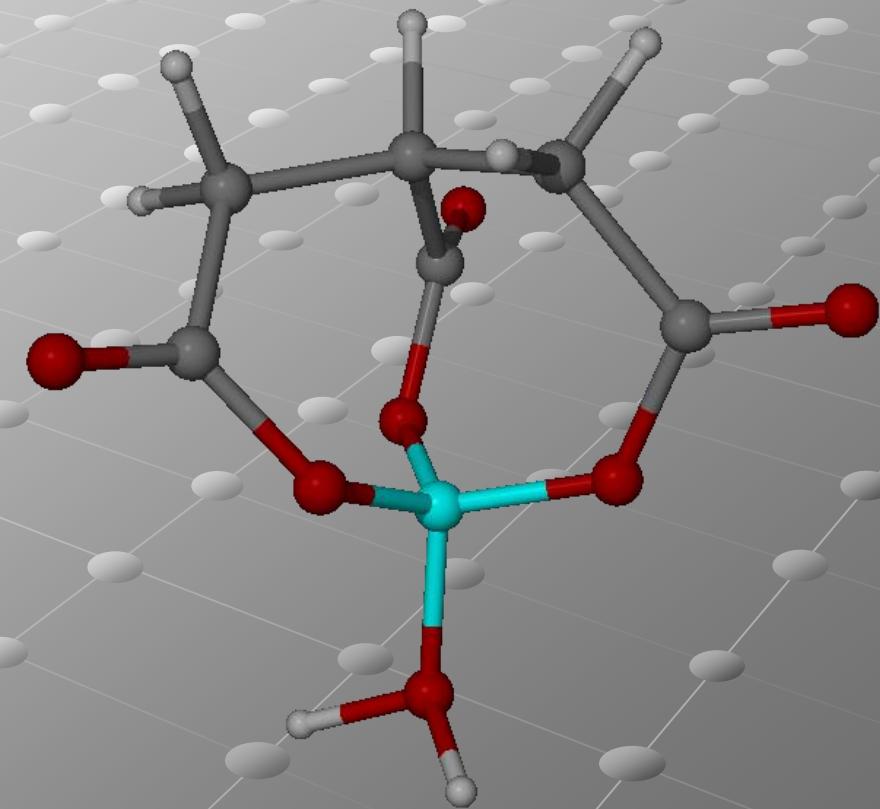
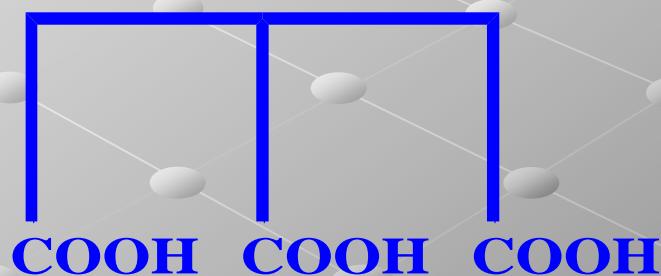
*Laboratory of Inorganic Chemistry
Department of Chemistry,
University of Ioannina, Greece*

MOFs

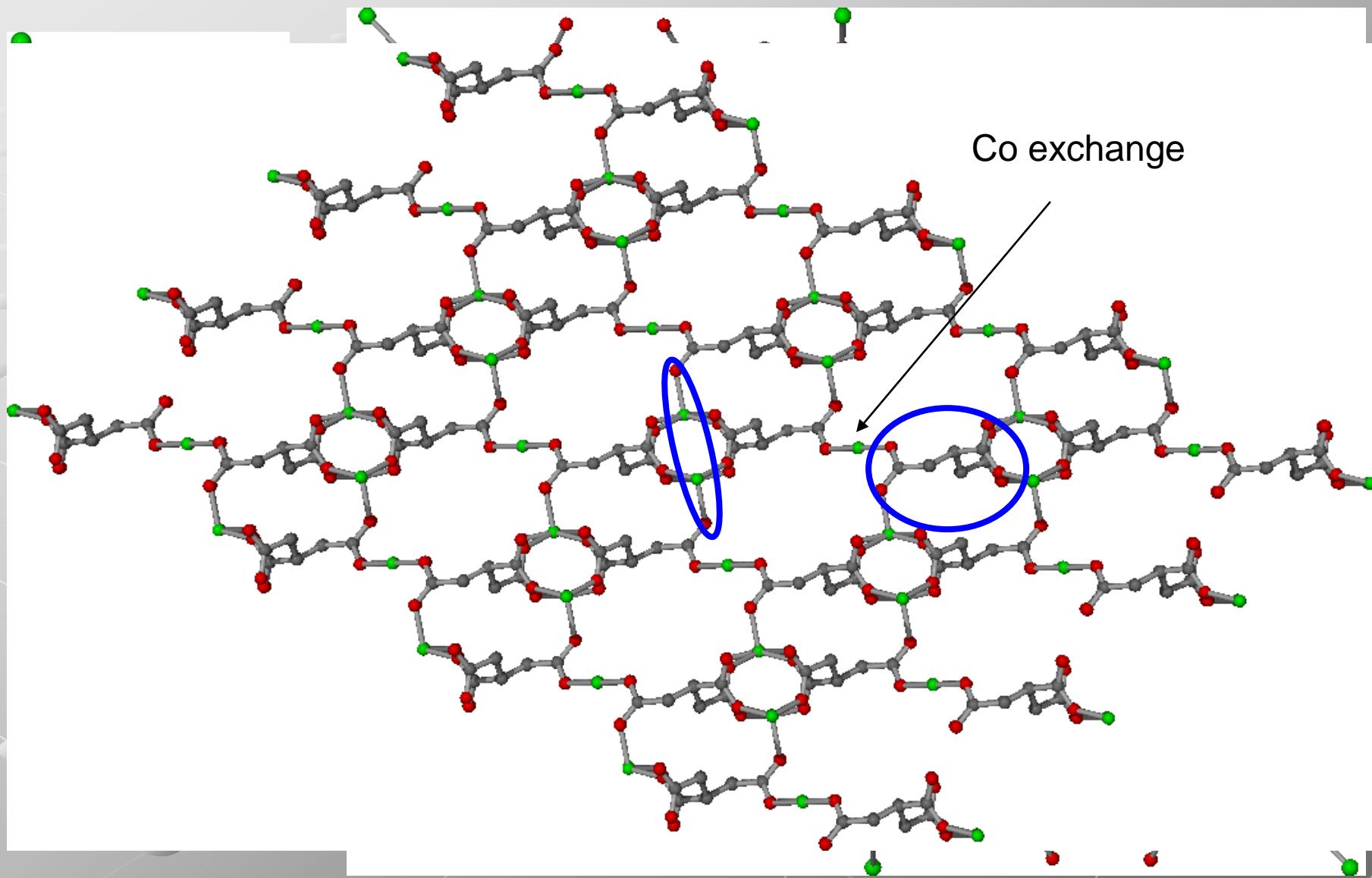


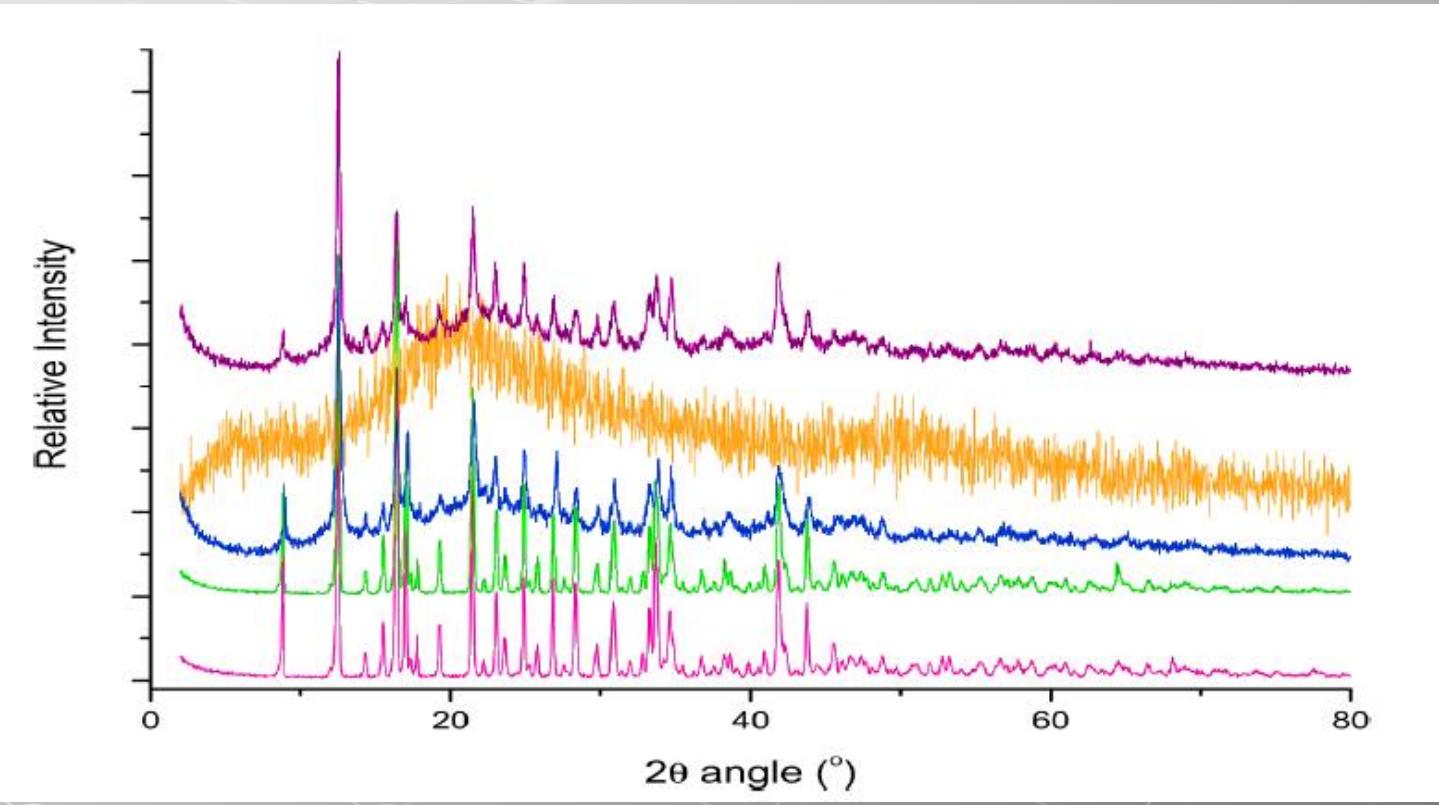
Bioinorganic Chemistry

Tricarballylic acid
1,2,3-propanetricarboxylic acid



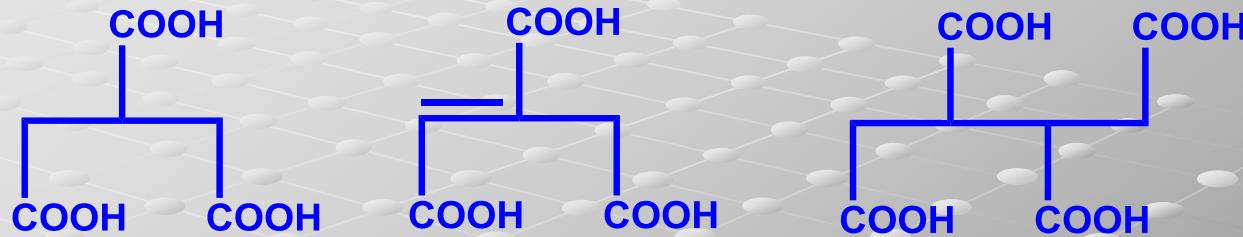
Zn^{2+} / PTC³⁻, $[\text{Zn}_3(\text{PTC})_2(\text{H}_2\text{O})_4]_n \cdot 2n\text{H}_2\text{O}$, 3D polymer, hexadentate
 μ_6 - carboxylate



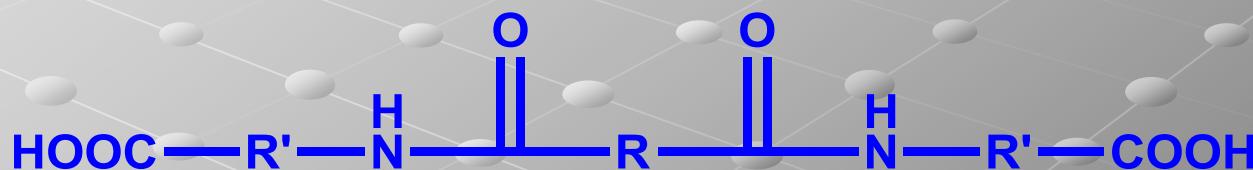


Ligand Families

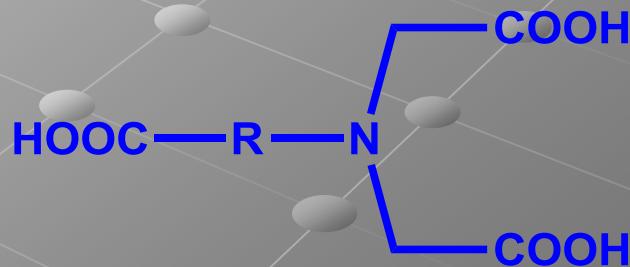
- Open Chain Polycarboxylates



- Pseudopeptides



- Iminodiacetates



a. Open chain polycarboxylates

1,2,3-propanetricarboxylic acid

PTCH_3

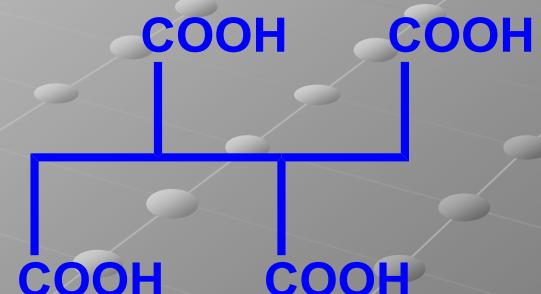
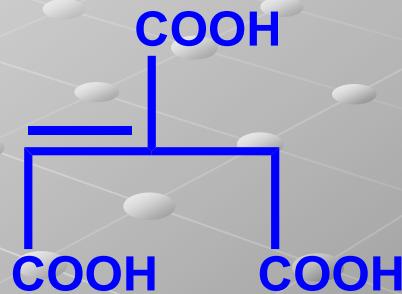
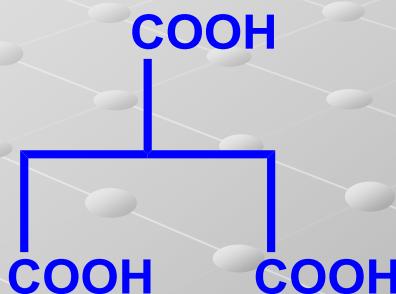
1,2,3,4-butanetetracarboxylic acid

BTCH_4

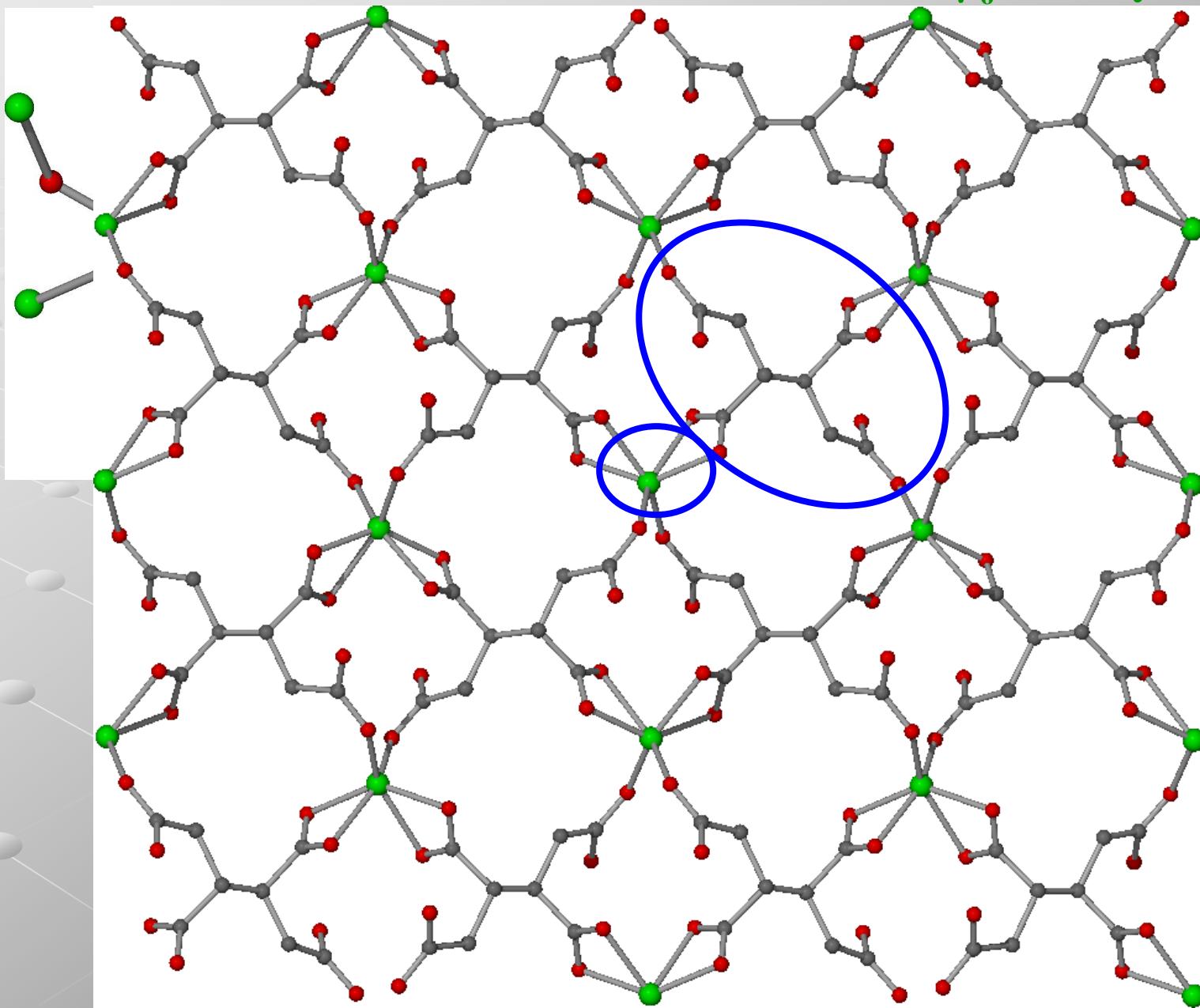
trans-1,2,3-propenetricarboxylic acid

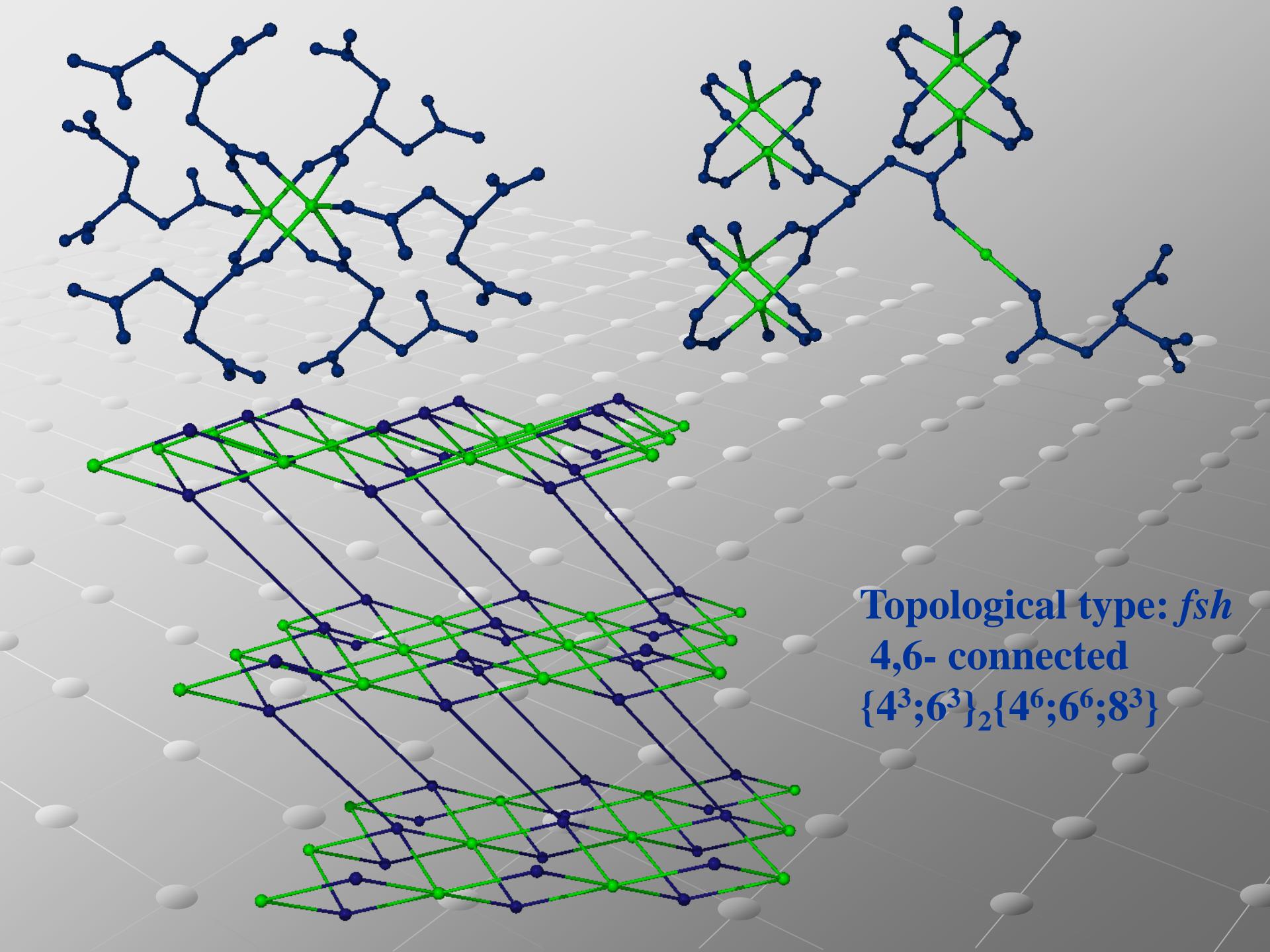
TAAH_3

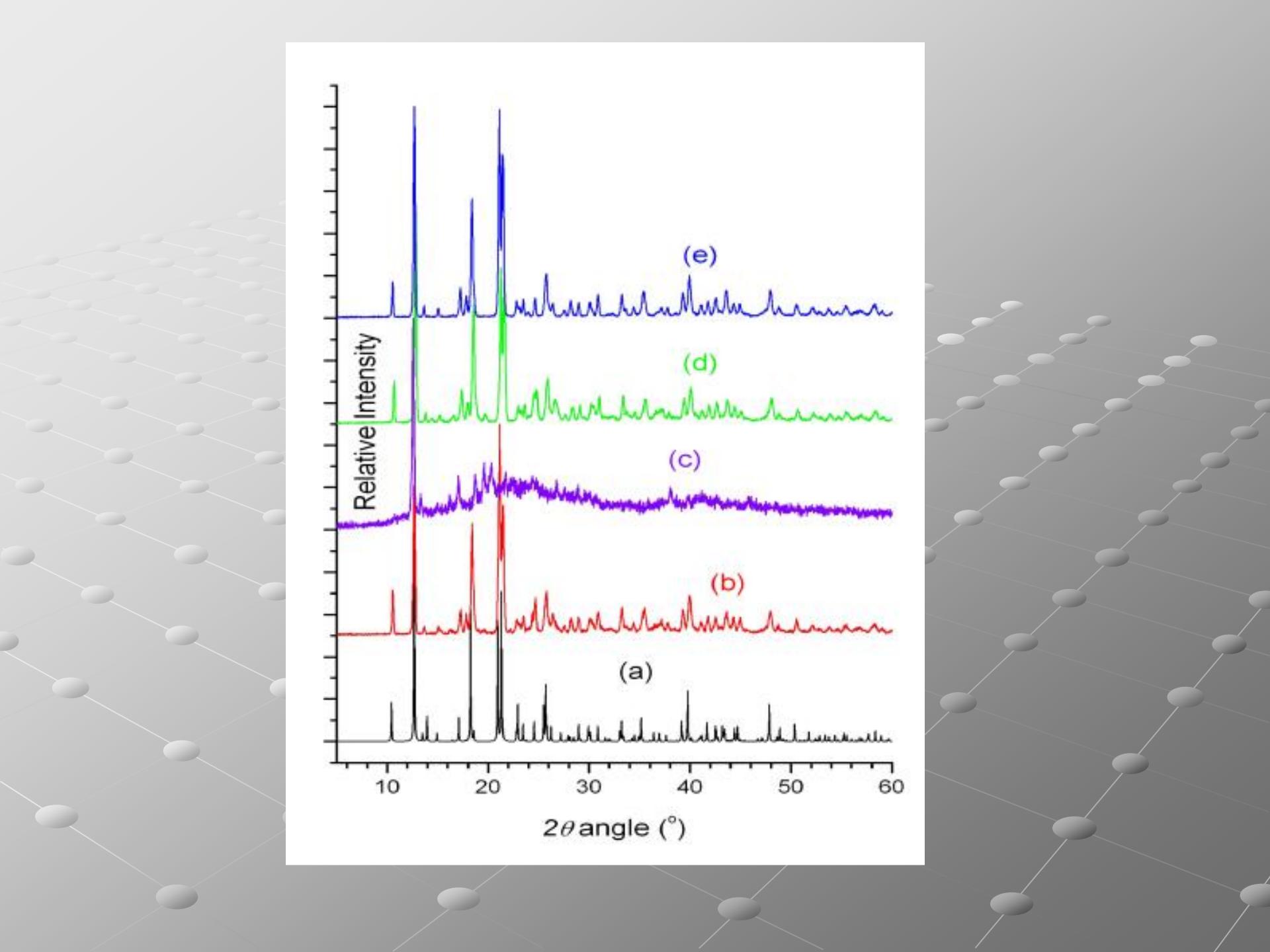
+ Lewis bases



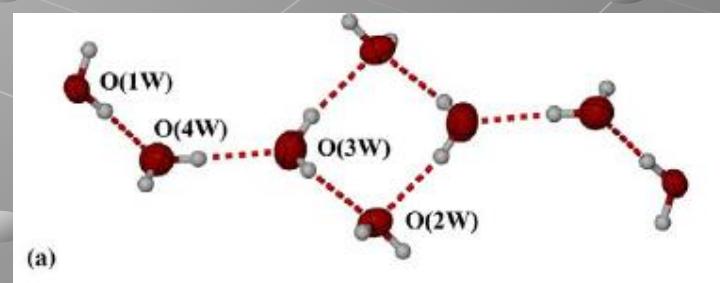
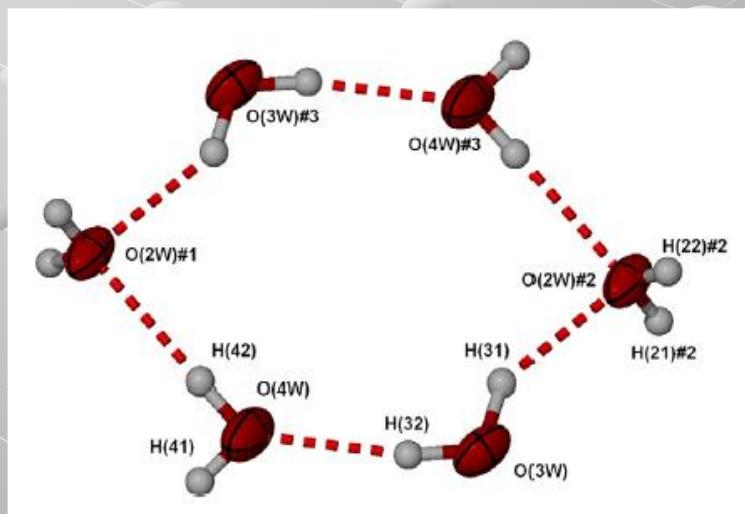
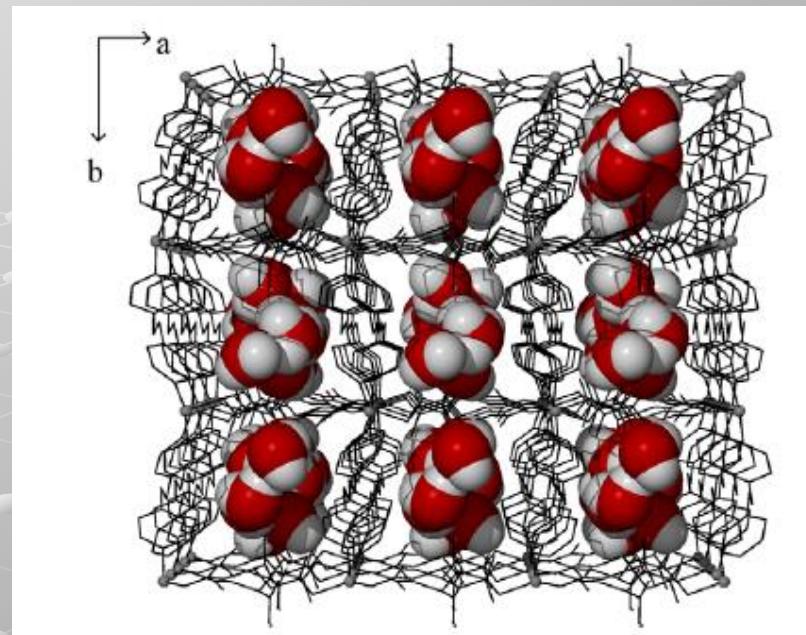
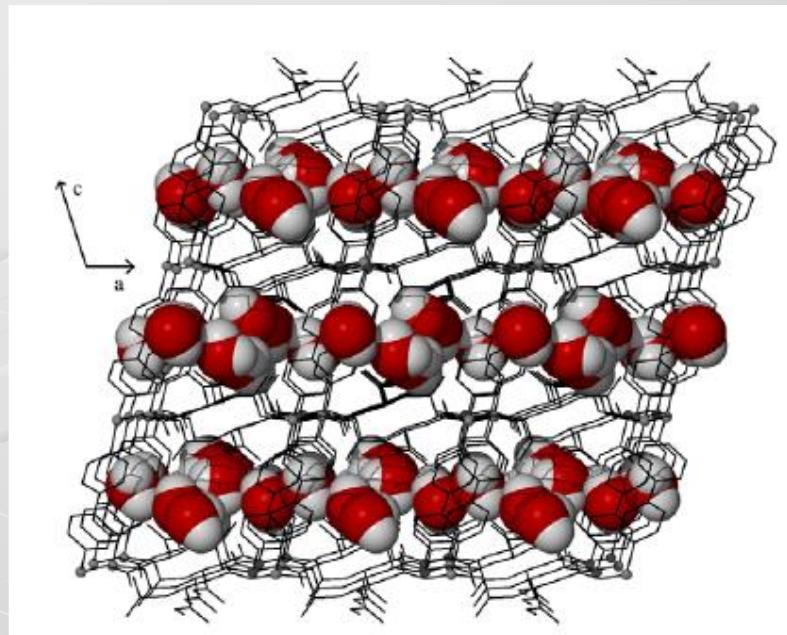
Zn^{2+} / BTC^{4-} , $[\text{Zn}_2(\text{BTC})(\text{H}_2\text{O})_4]_n \cdot 2n\text{H}_2\text{O}$, 3D polymer, Octadentate
 μ_6 -carboxylate



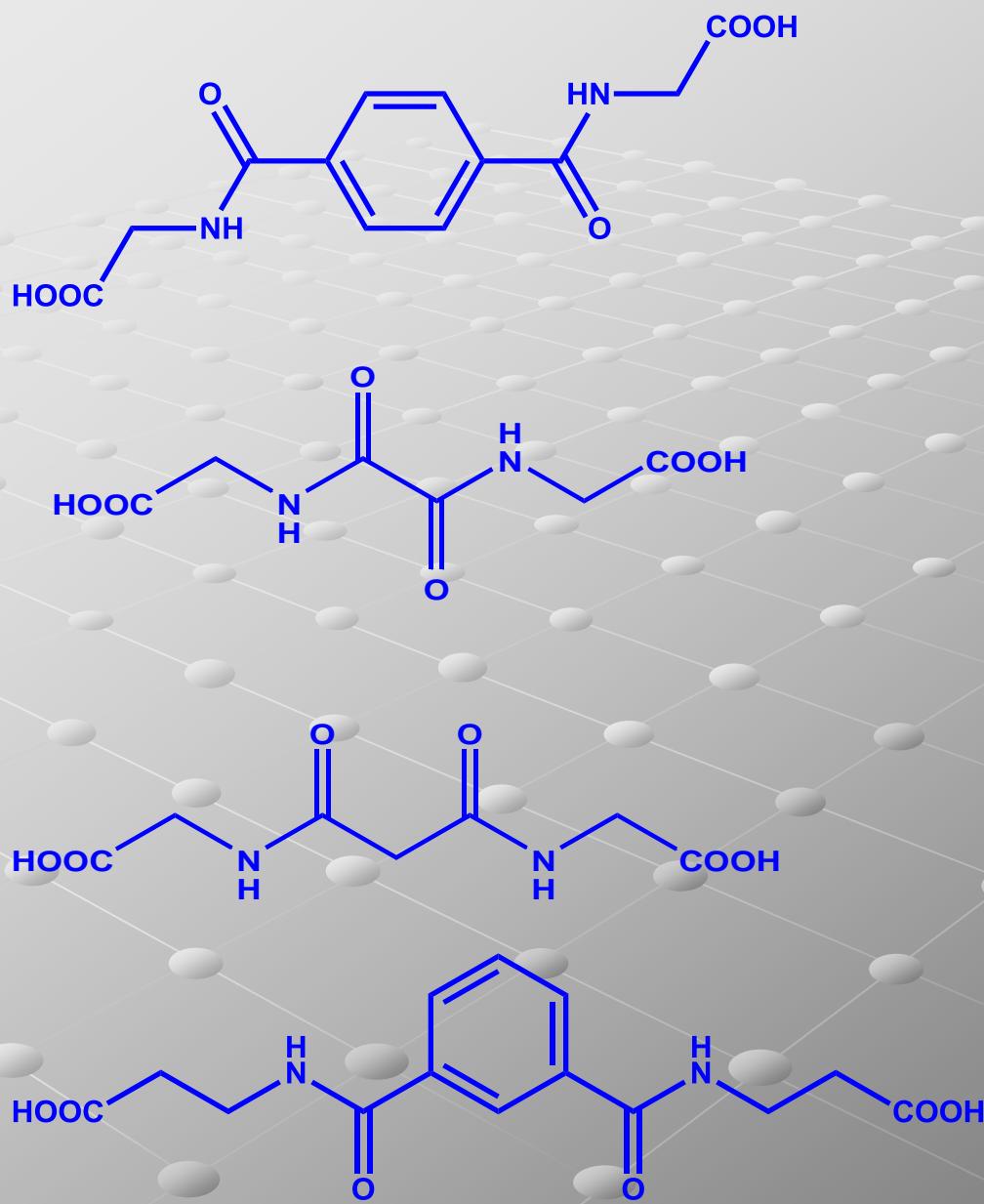




Addition of Lewis bases in metal ion – polycarboxylate reaction mixtures



b. Pseudopeptides



TBGH₂

OBGH₂

MBGH₂

IB β AH₂

TBG²⁻

Co²⁺

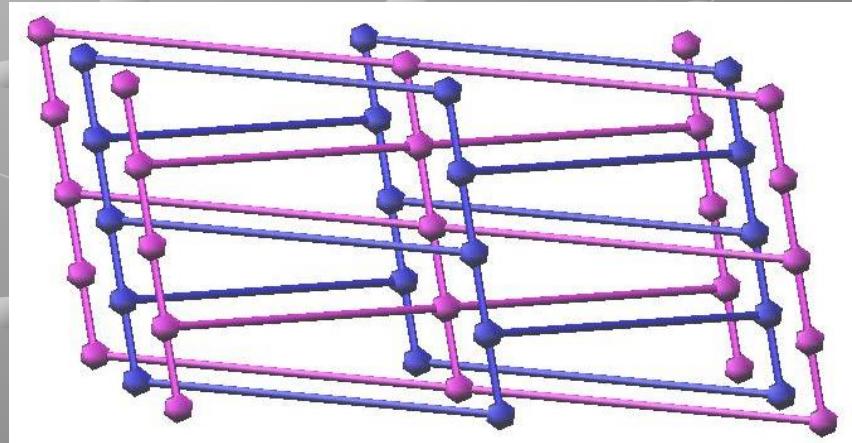
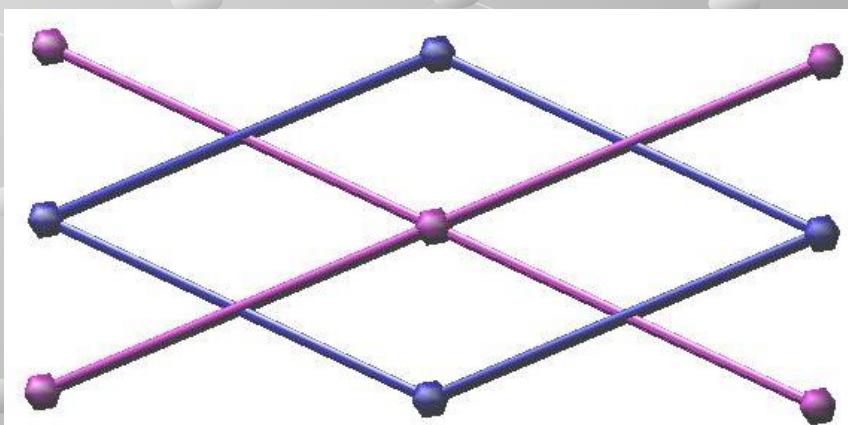
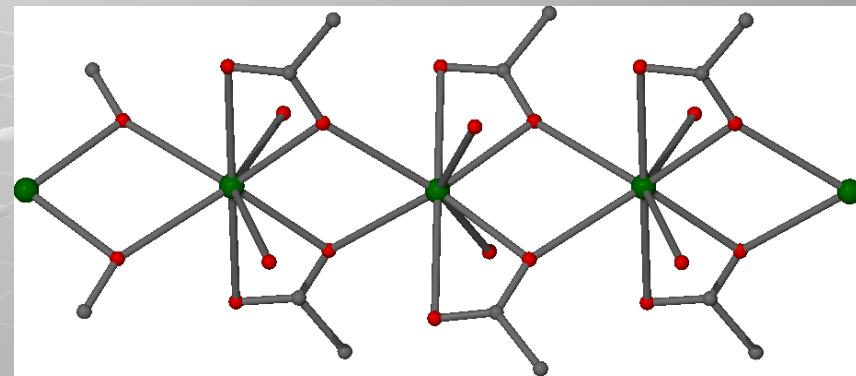
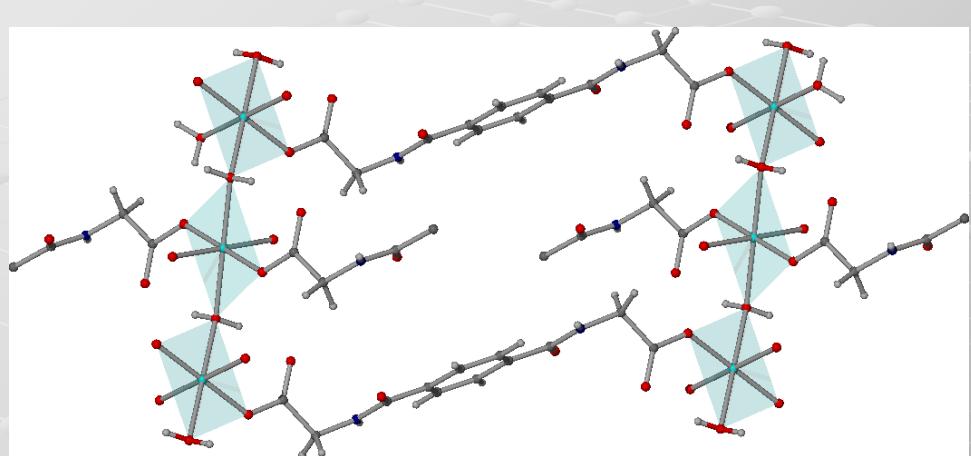
Cu²⁺ Interpenetrated 3D polymers,

Pb²⁺

μ_2 - carboxylate

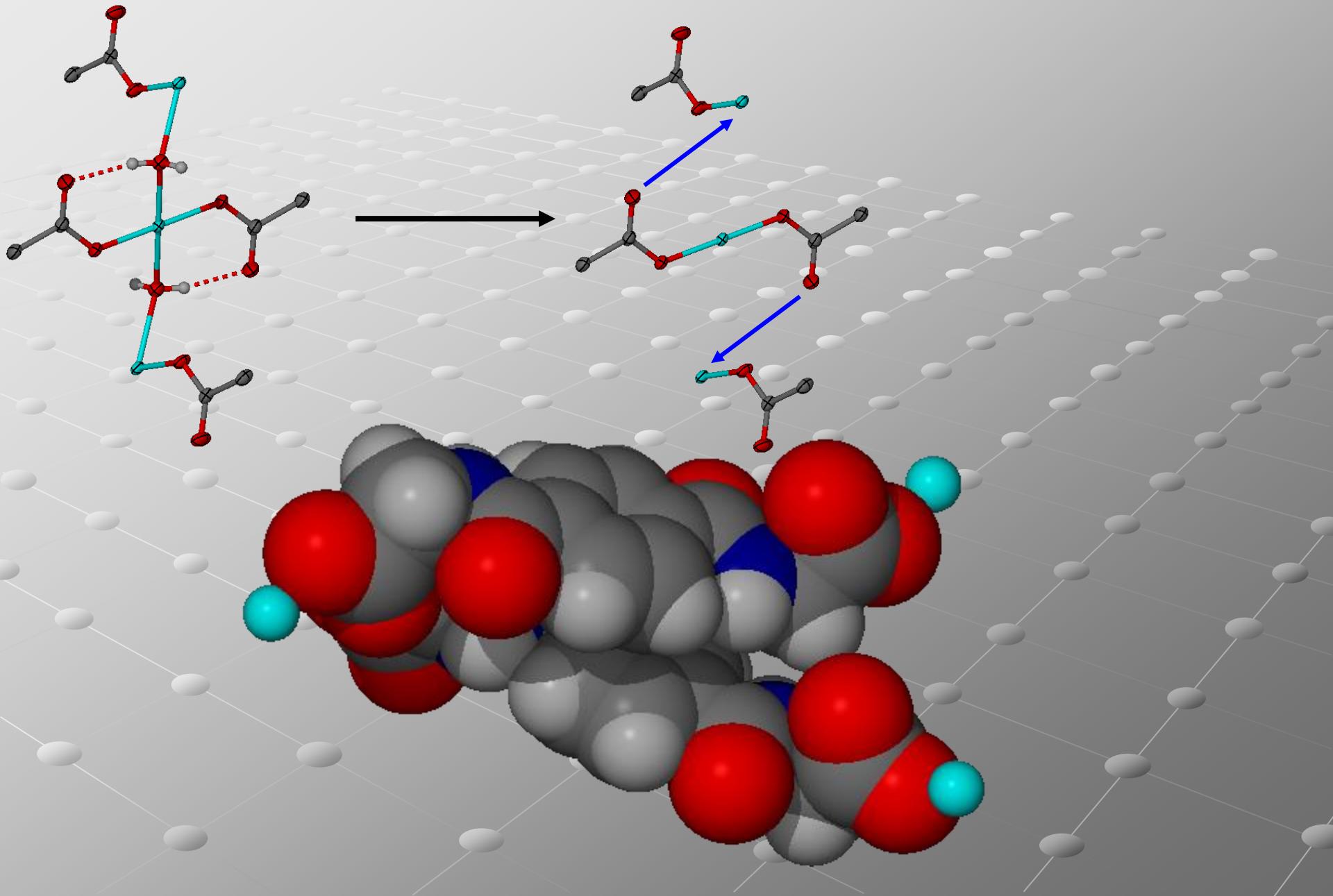
μ_2 - carboxylate

Hexadentate μ_4 - carboxylate

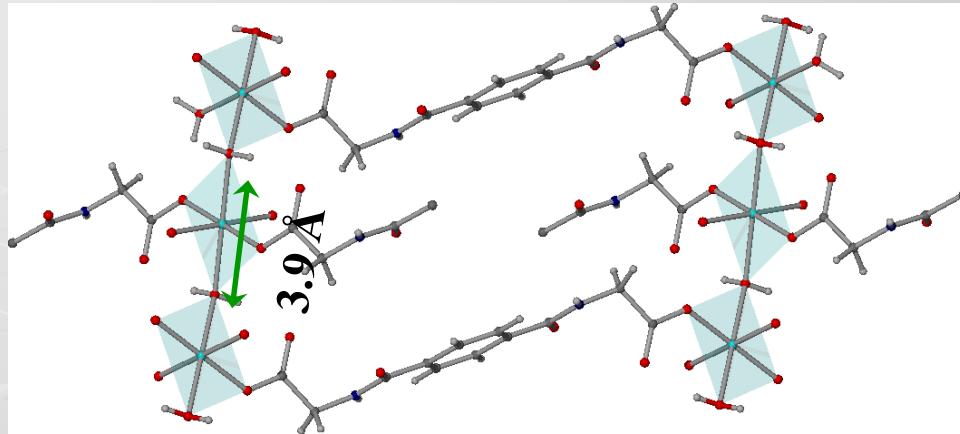


Topological type: cds, 4 – connected {6⁵;8}

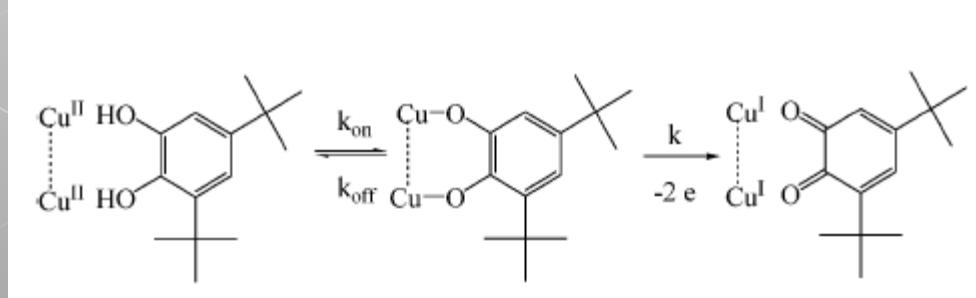
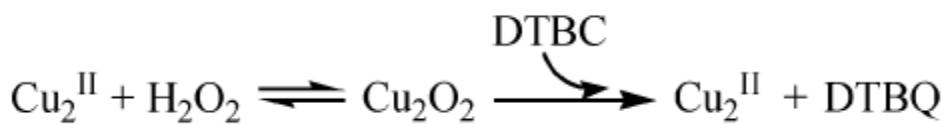
Reversibility of water loss



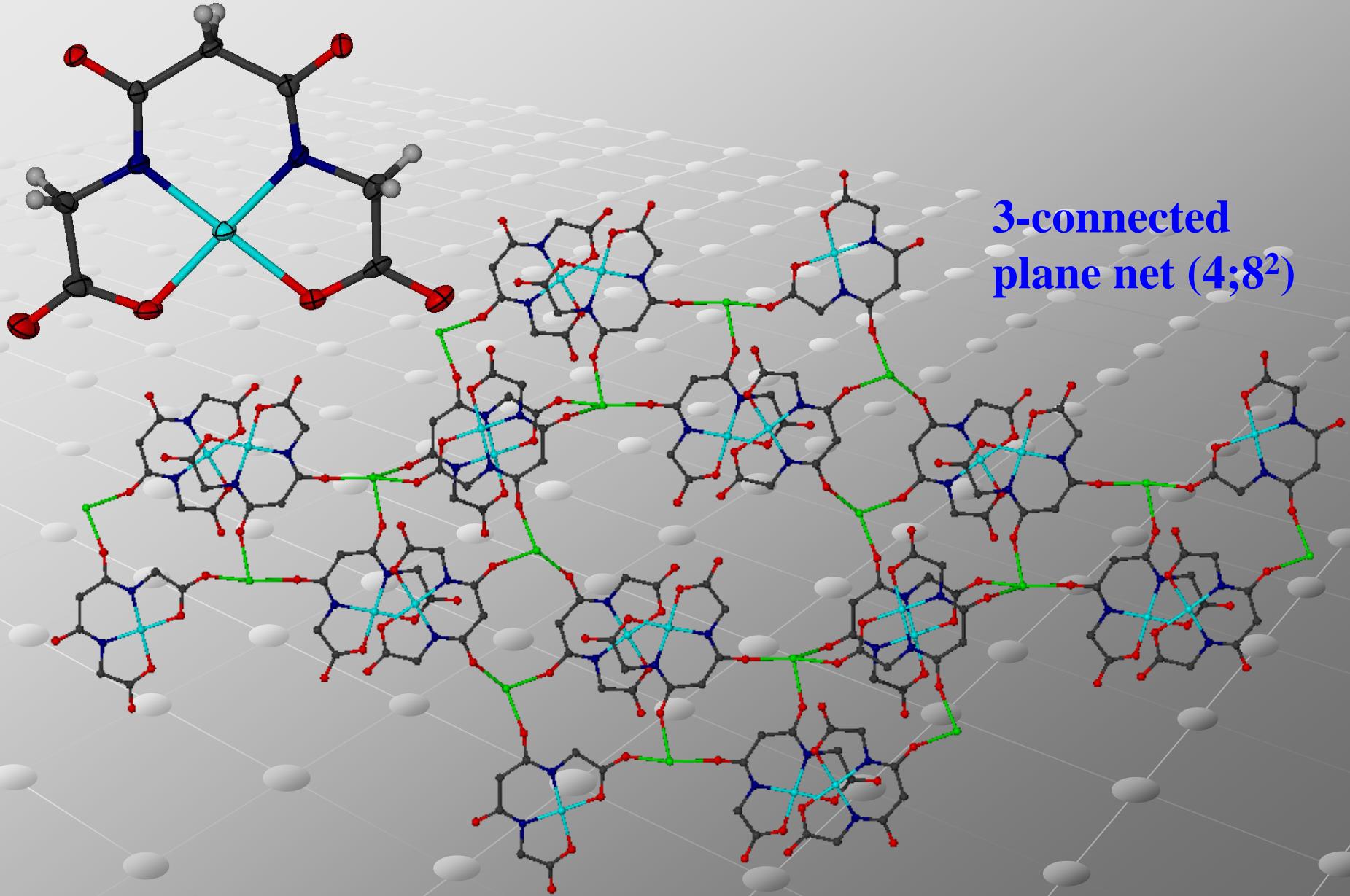
Catalytic Properties of the Cu^{II}-TBG²⁻ polymer



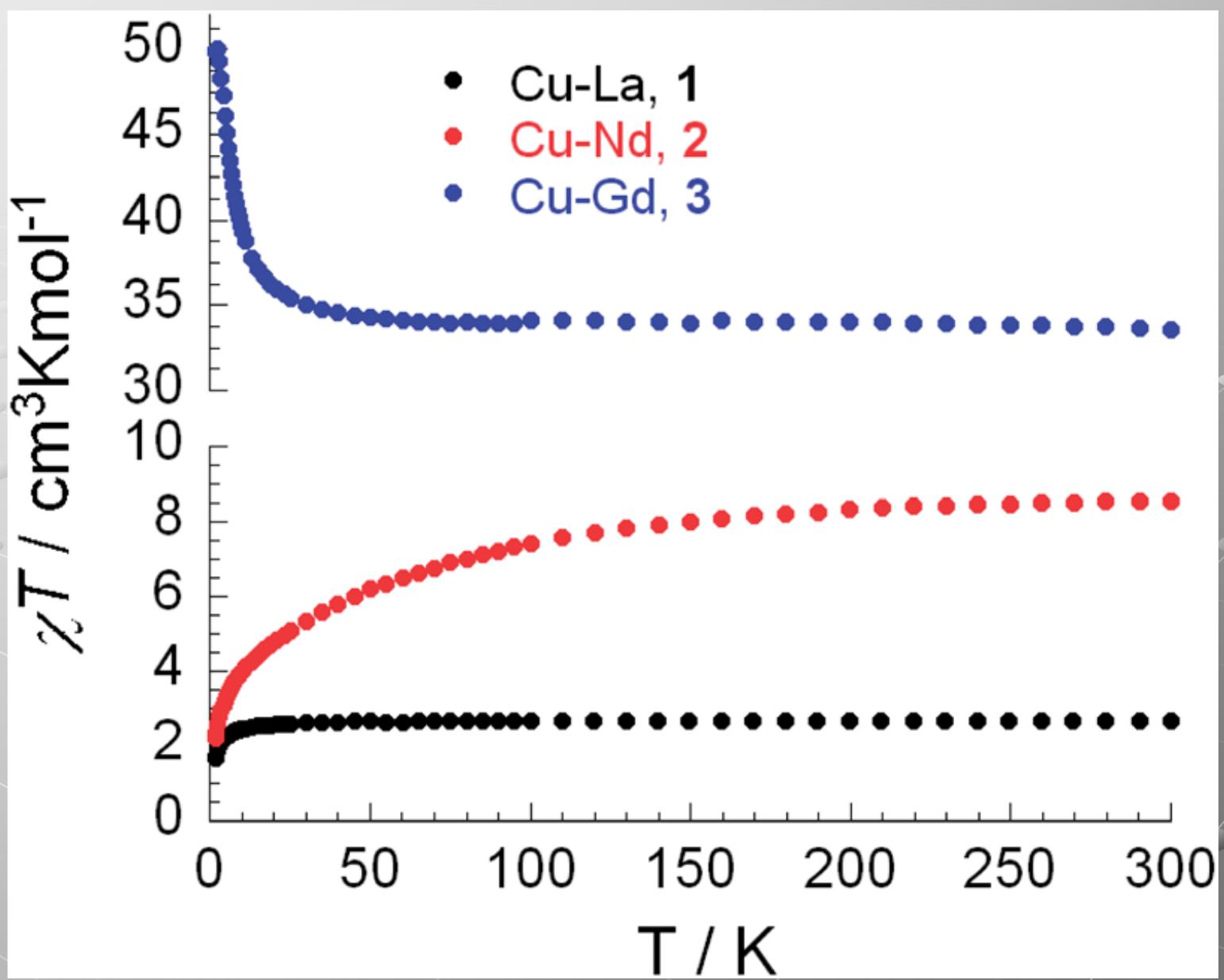
Resembles tyrosinase active site
and catalyses the oxidation of DTBC
heterogeneously, in the presence of H₂O₂



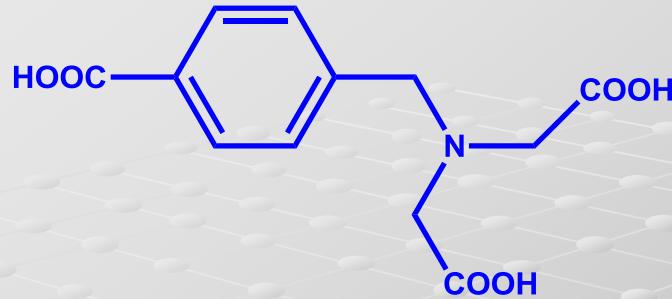
Malonylbisglycinate, MBG²⁻



3-connected
plane net (4;8²)



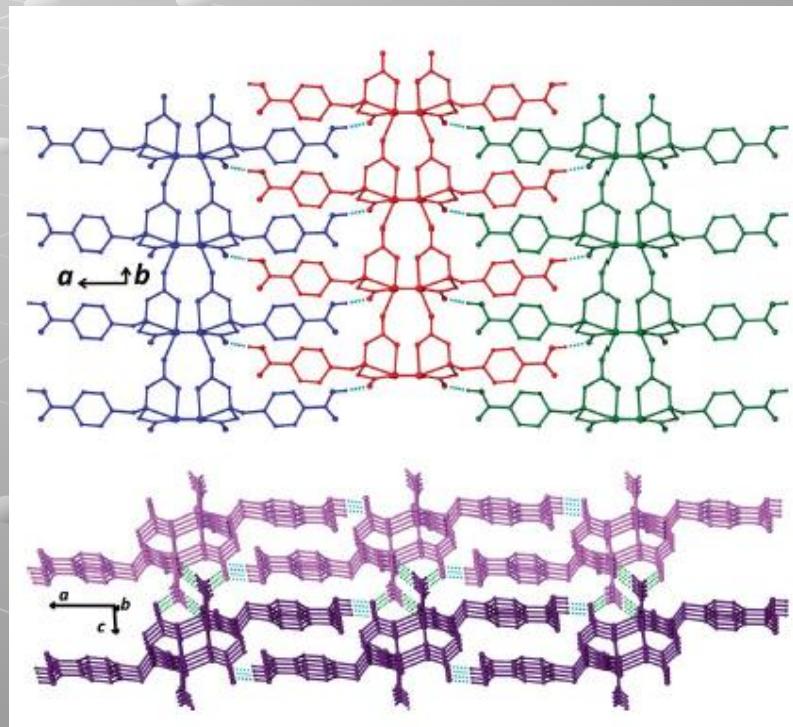
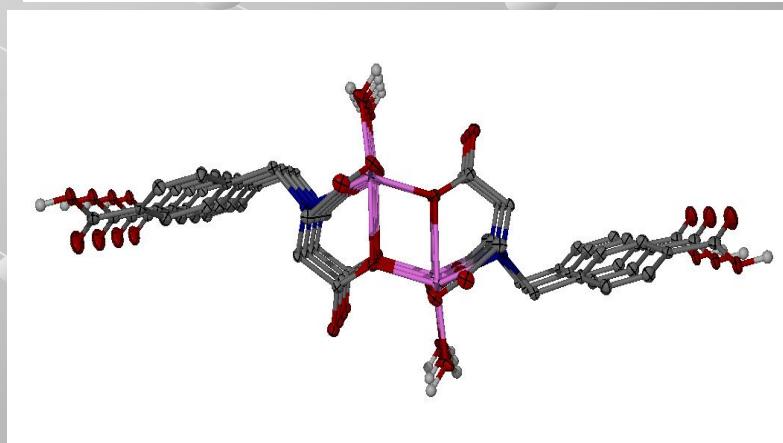
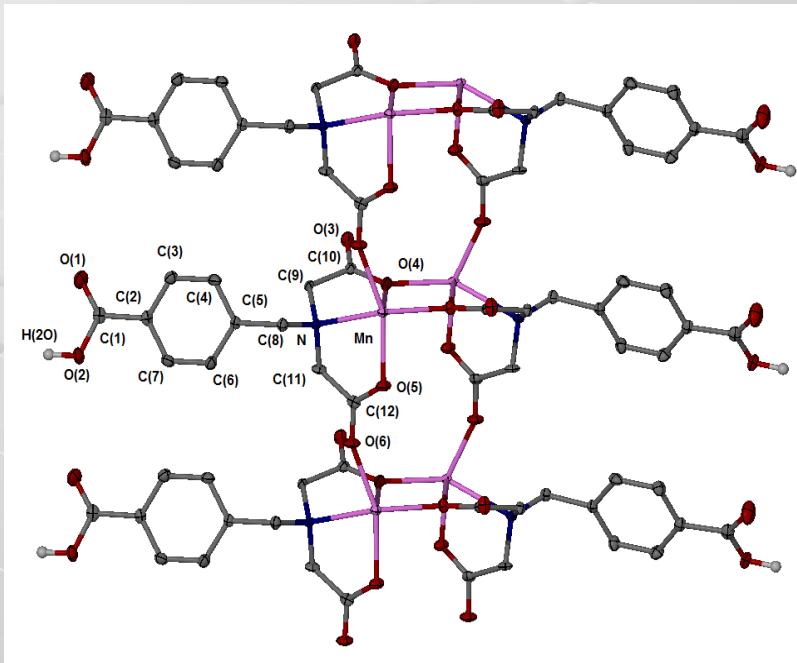
c. Iminodiacetates



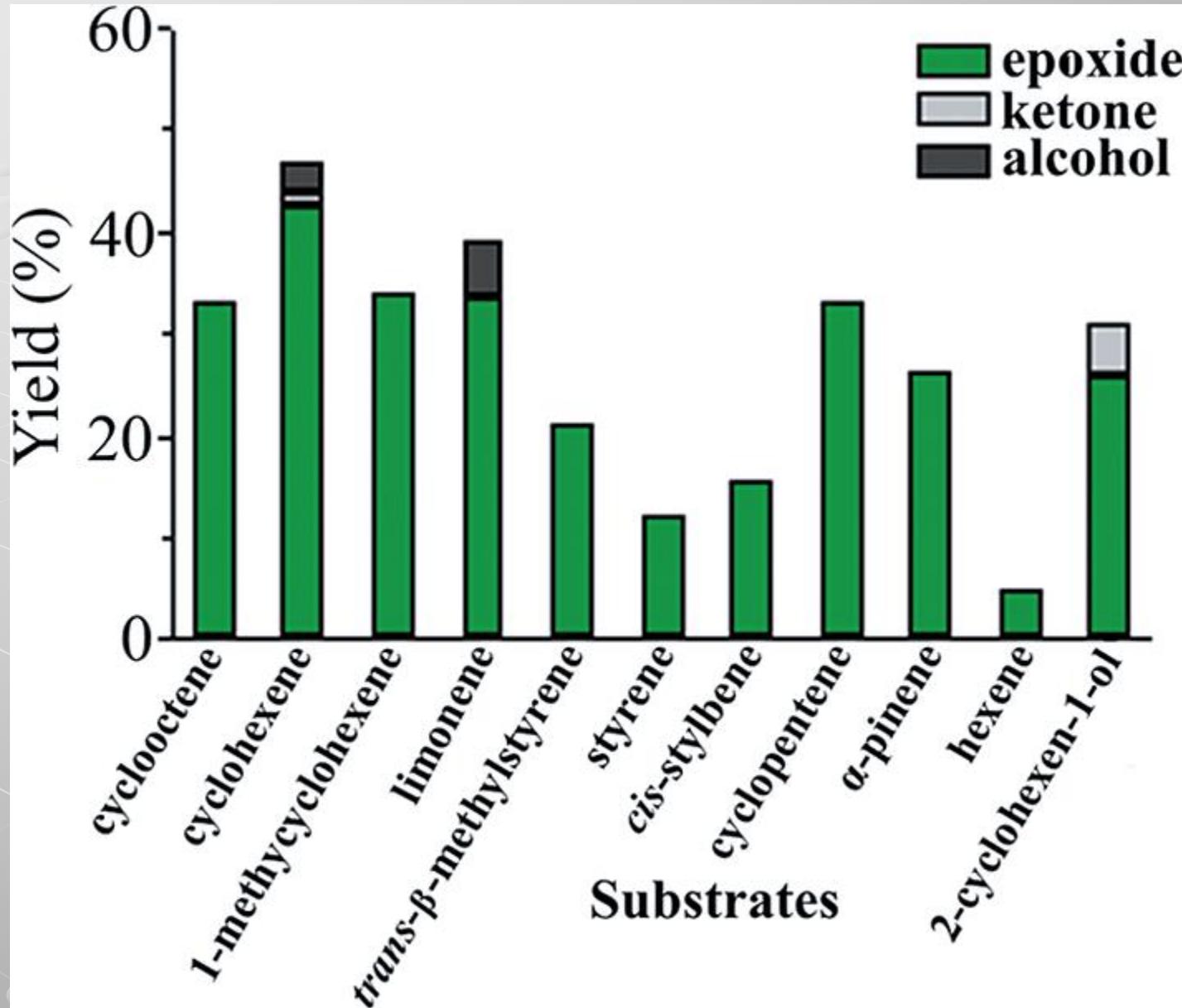
N-(4-carboxybenzyl)iminodiacetic acid

**Unprecedented coordinating behavior
Four different protonation states with ten different
coordination modes**

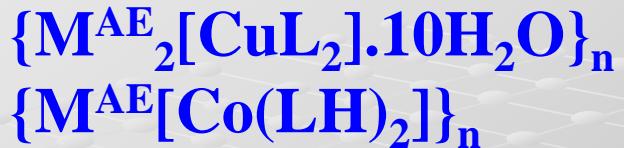
1D Polymers



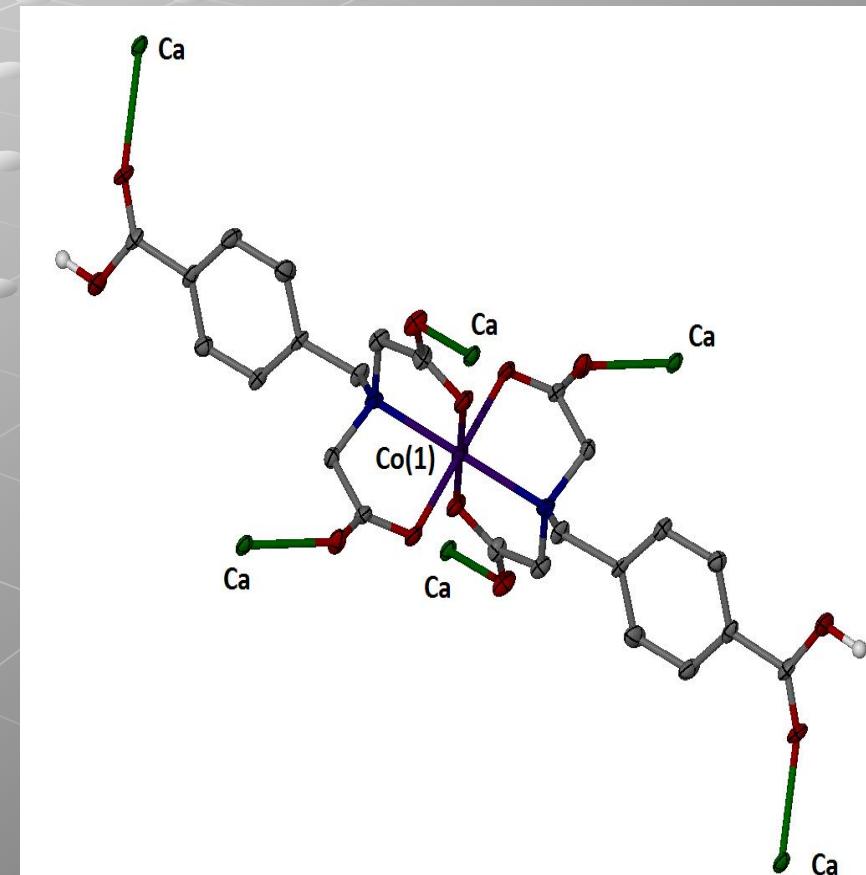
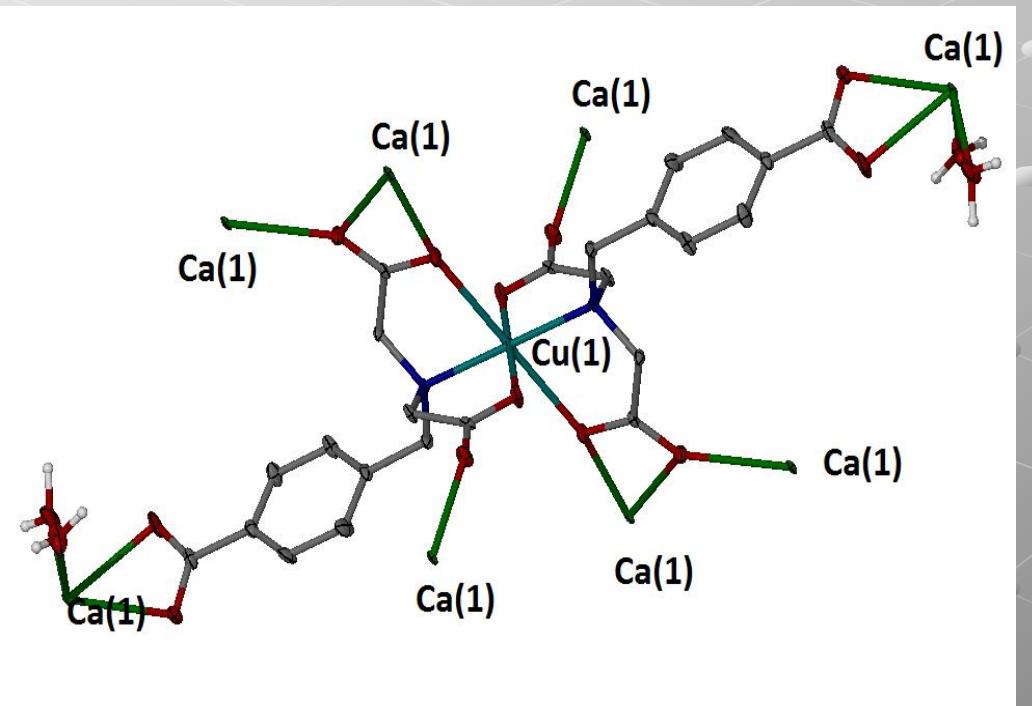
Heterogeneous Oxidation Catalysts

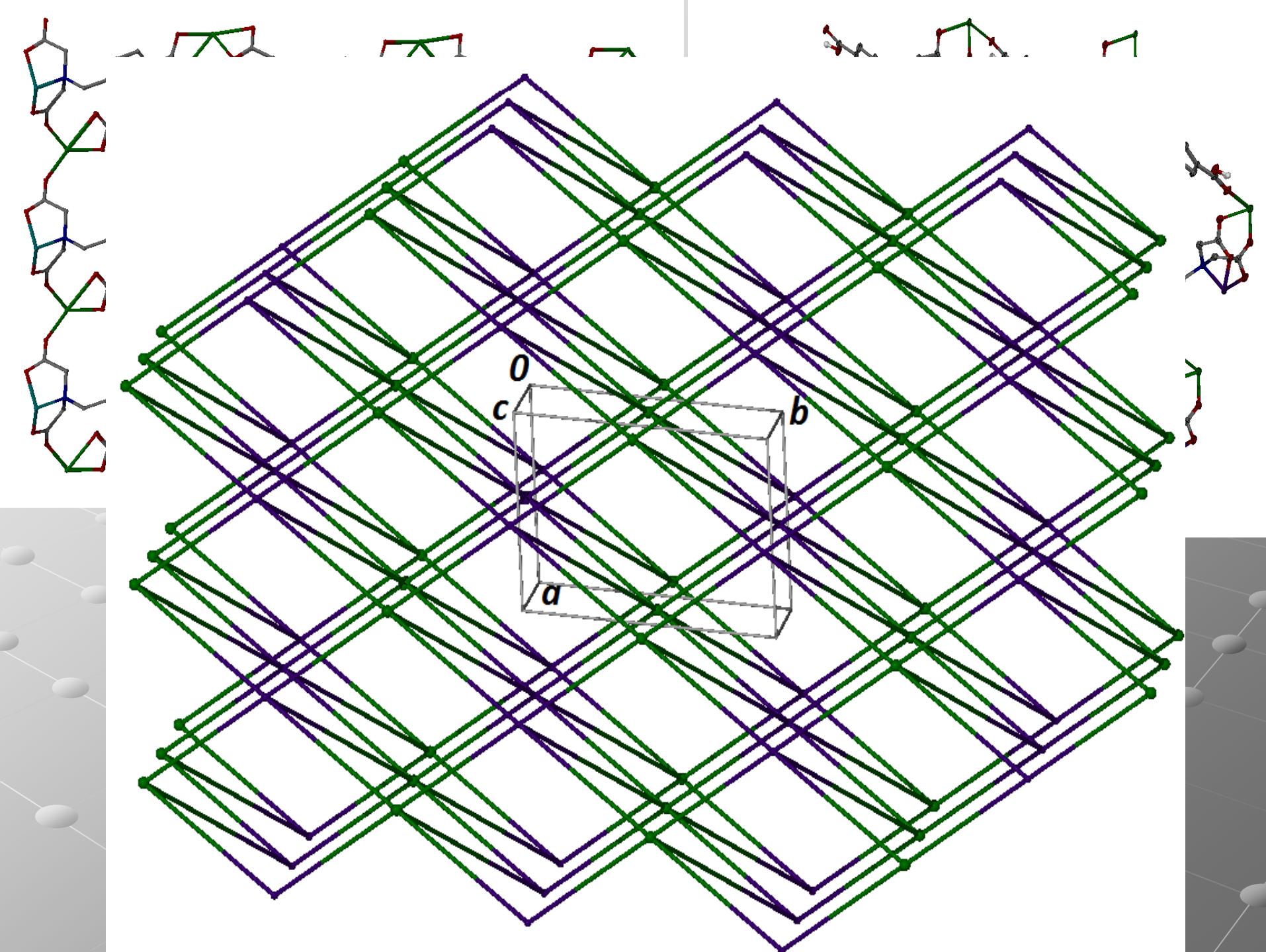


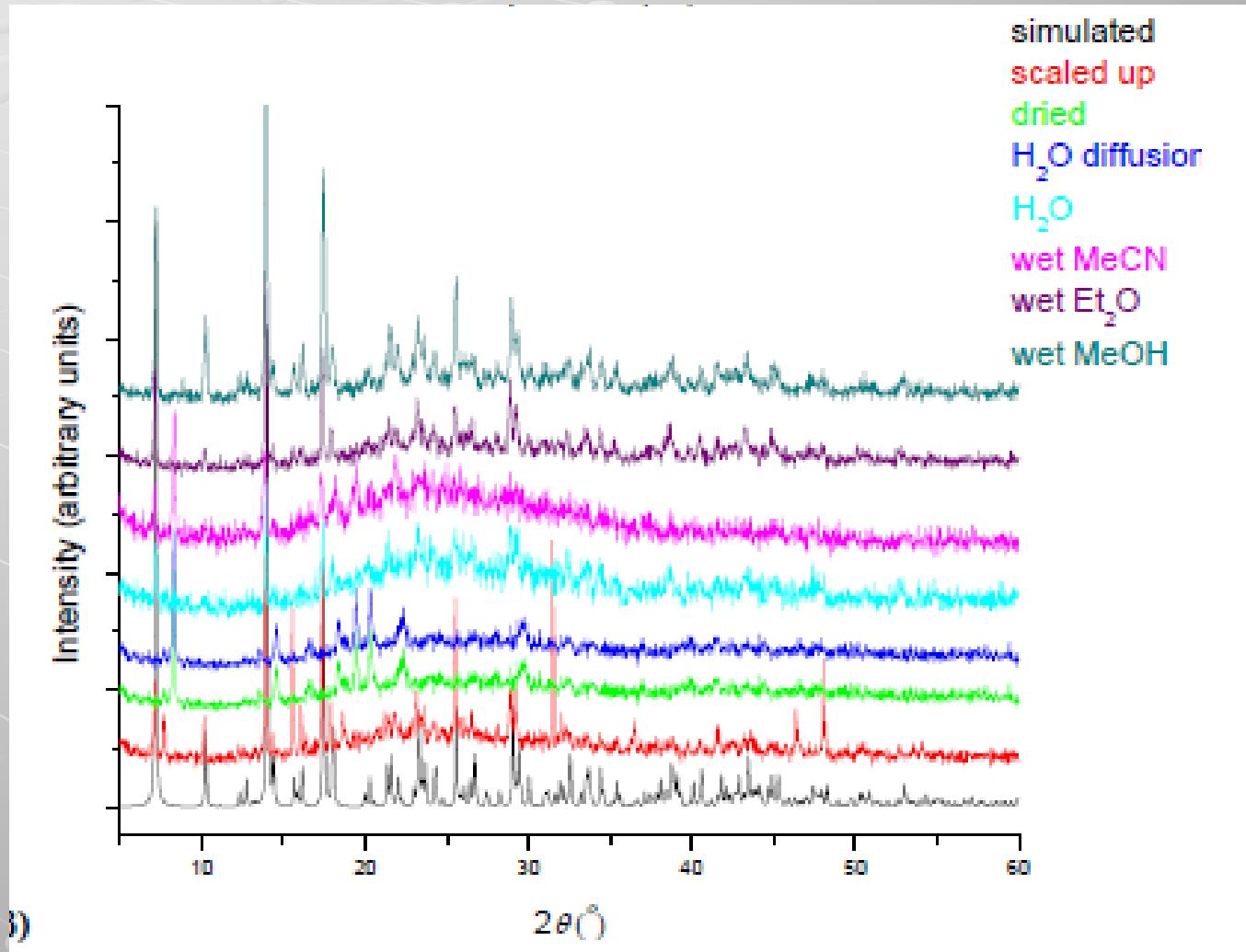
3D Mixed Metal Polymers



M = Ca, Sr, Ba

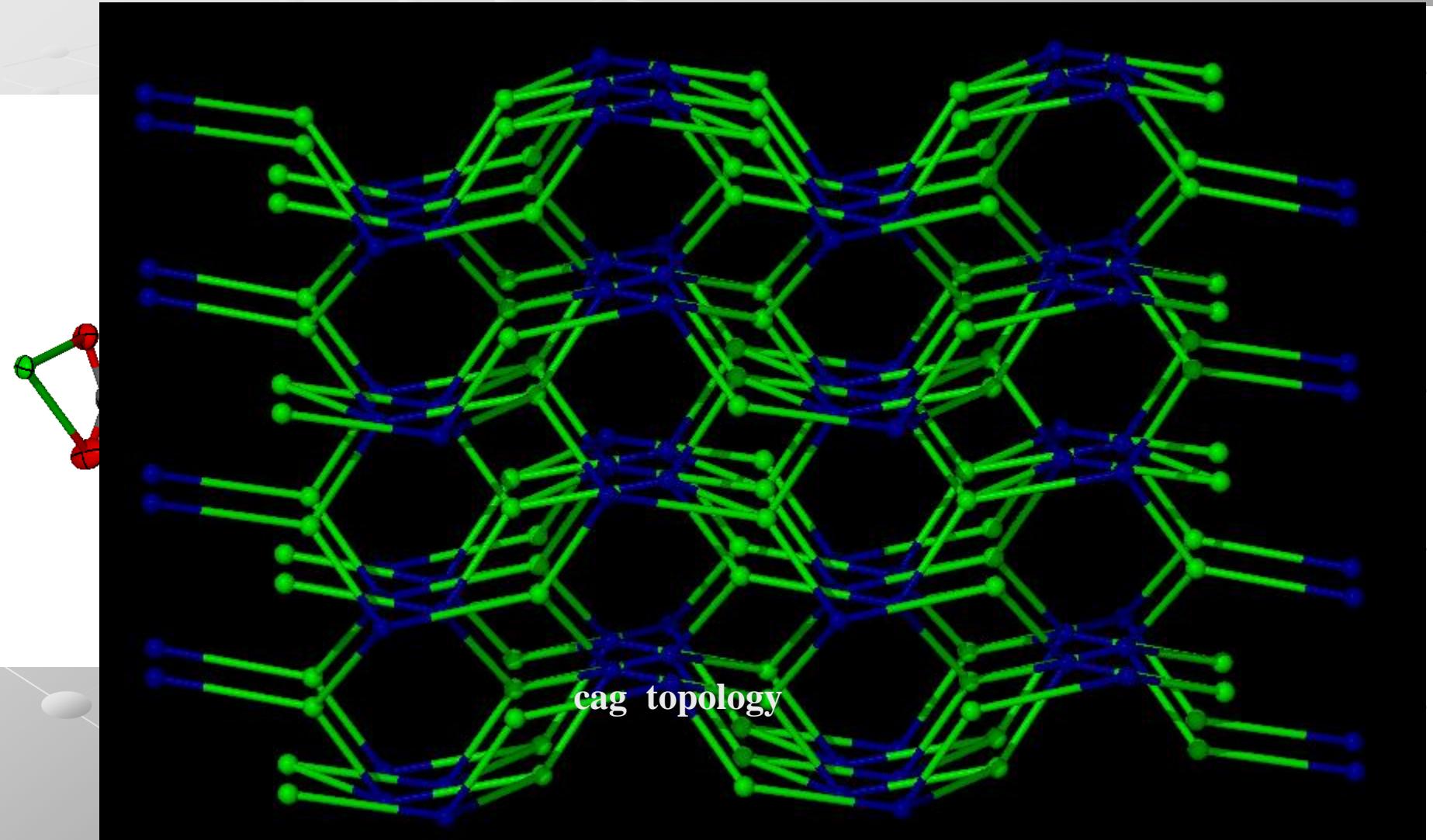


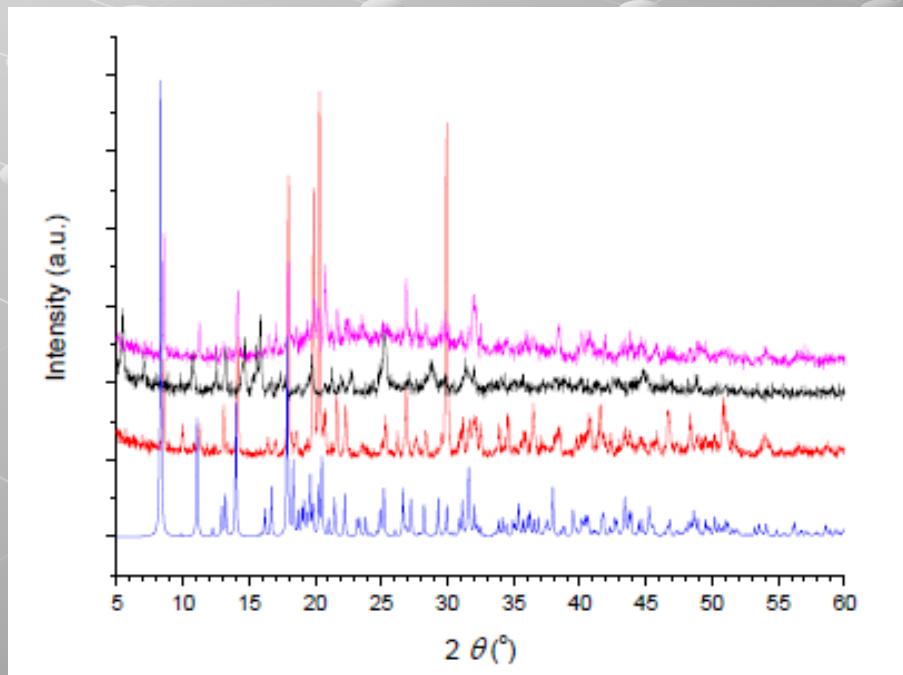
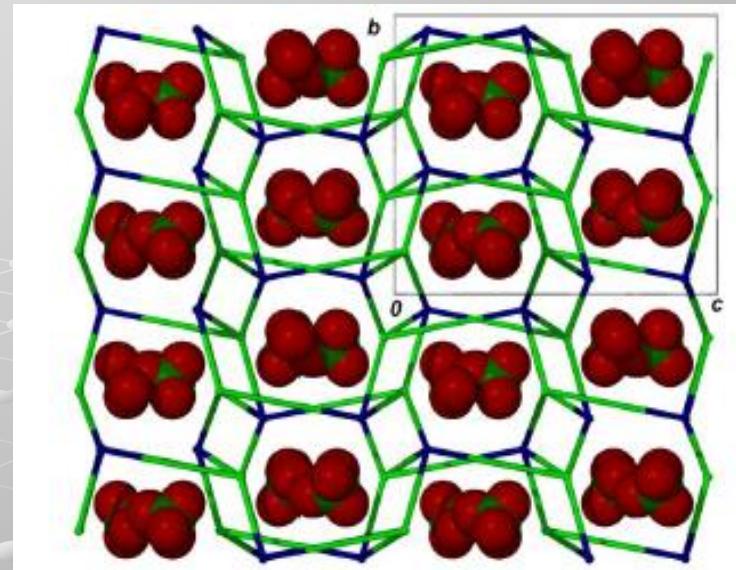
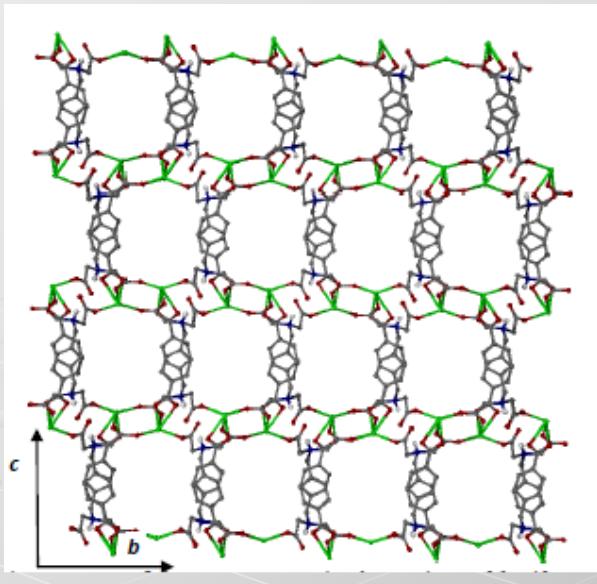




3D Cationic Polymers

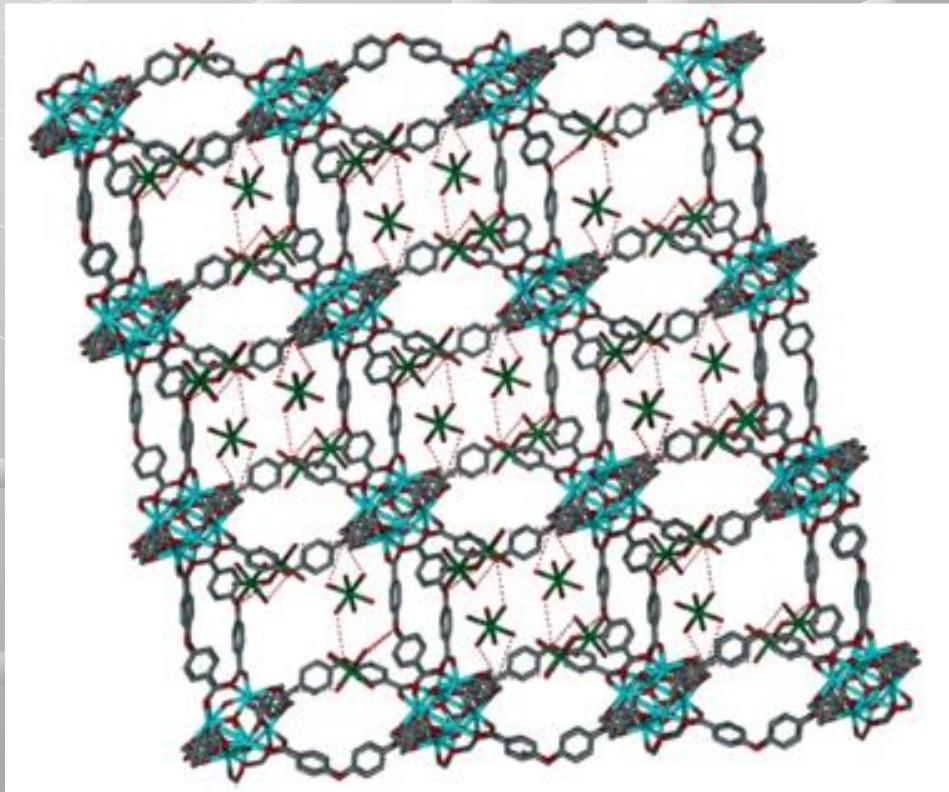
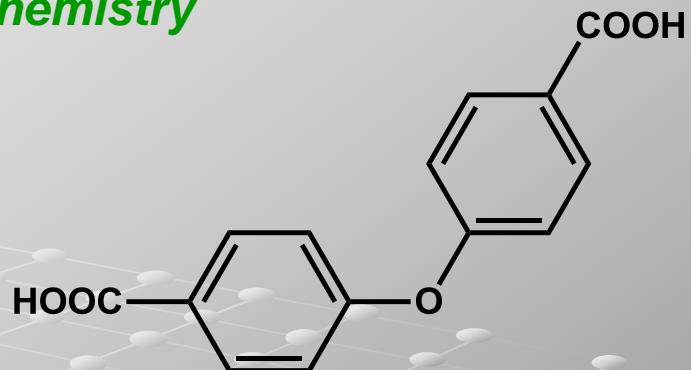
$\{[M(LH)(H_2O)_4]_n A_n\}$ M = La – Gd, A = NO_3^- , ClO_4^-



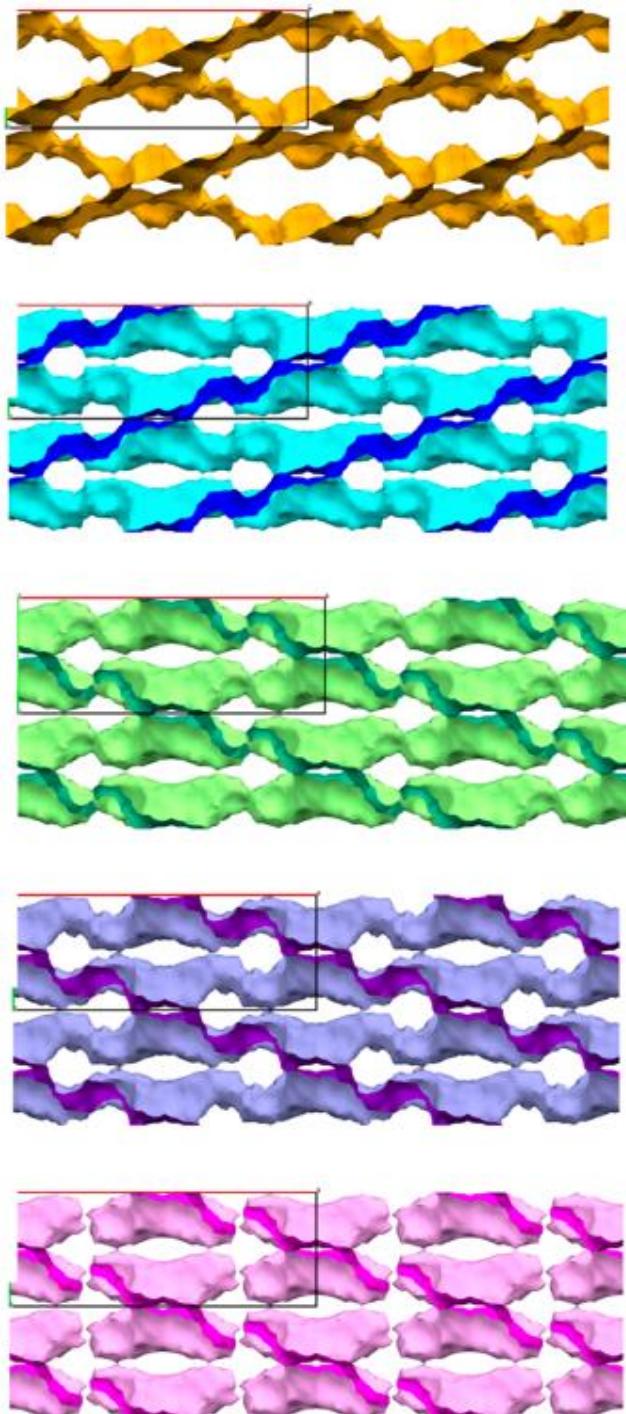
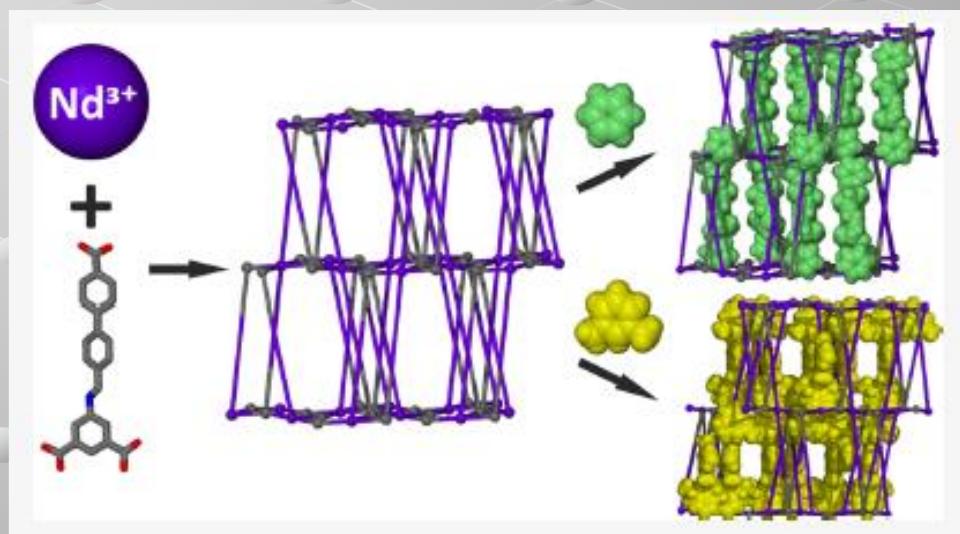
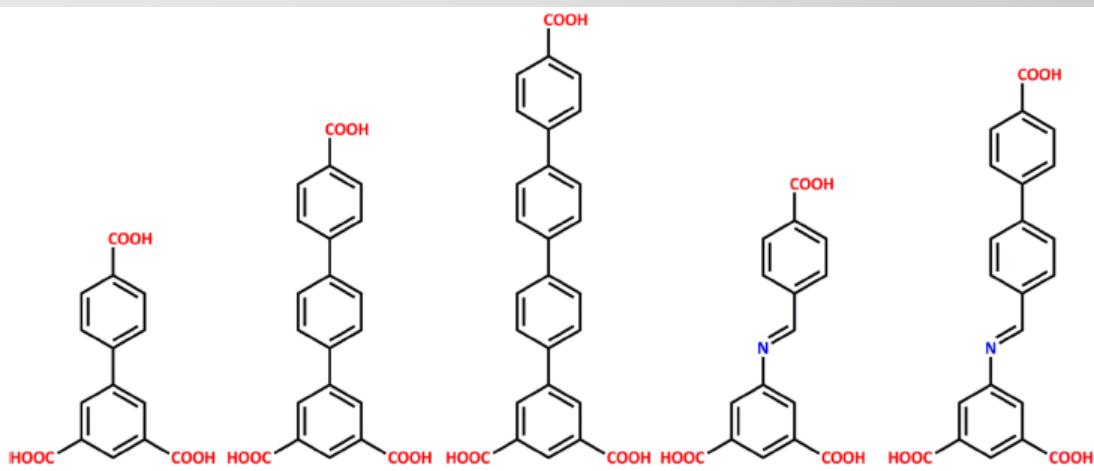


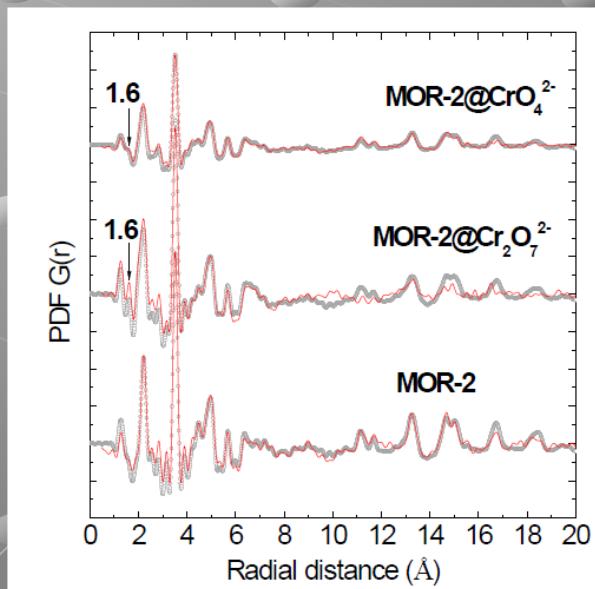
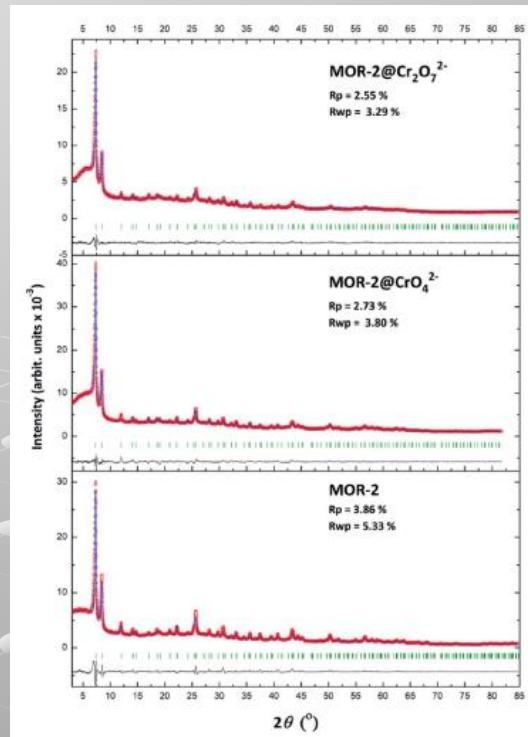
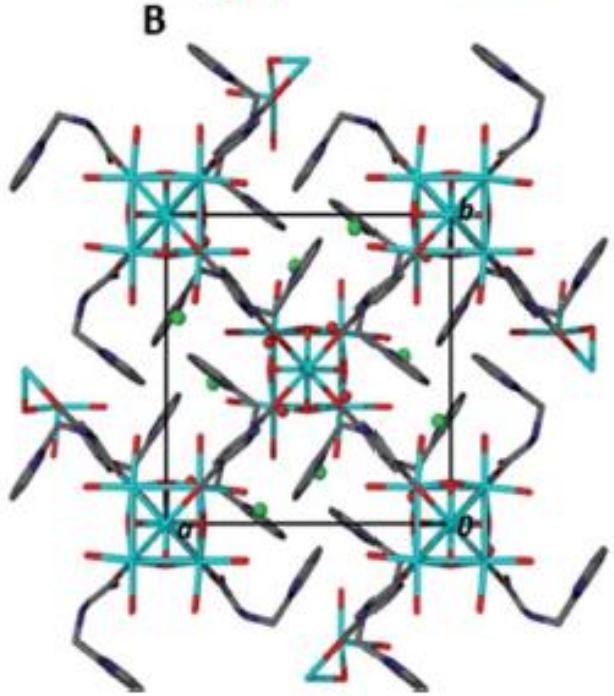
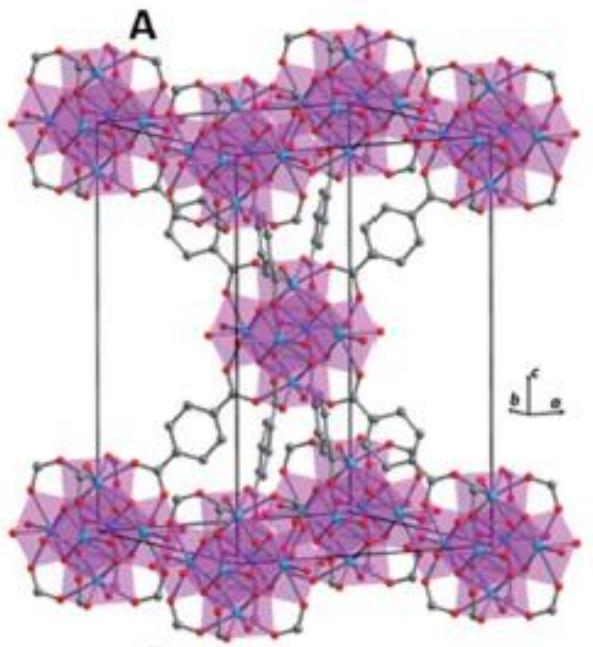
OBA Chemistry

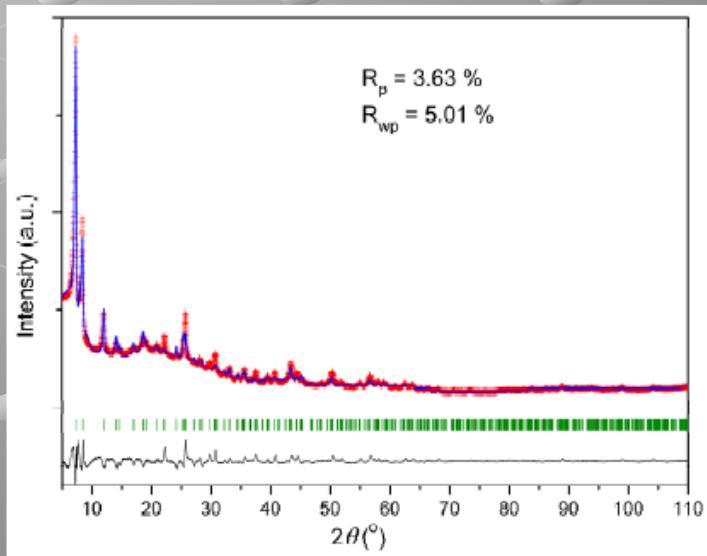
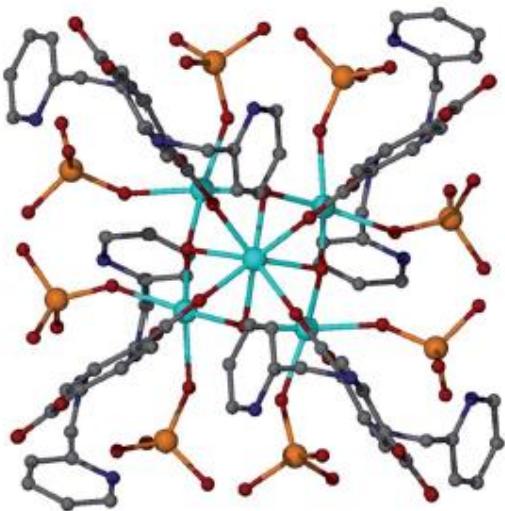
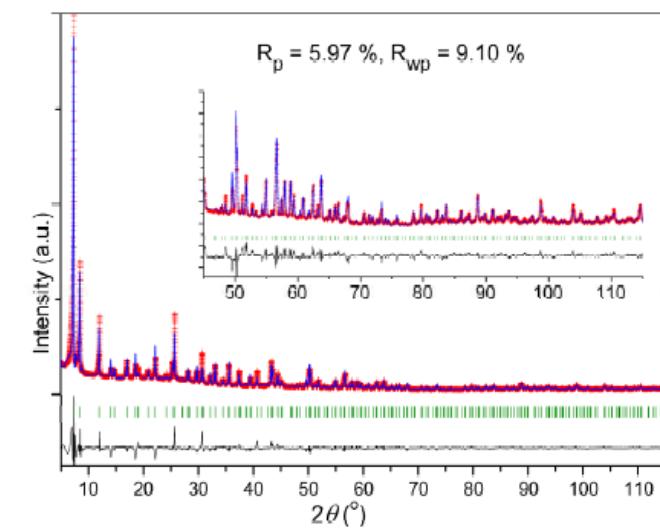
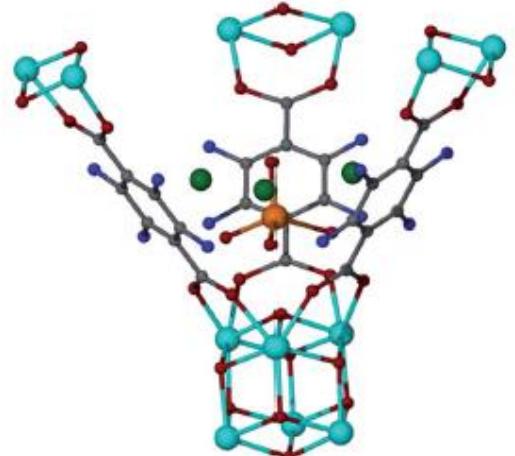
$[RE_6(\mu_3-OH)_8]^{10+}$



Isoreticular synthesis and SCSC transformations







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Prof. Adonis Michailidis

Prof. Stavroula Skoulika

Dr. Catherine Raptopoulou

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Dr. Athanassios Boudalis

Prof. Costas Milios

FINANCIAL SUPPORT



Ευρωπαϊκή Ένωση
Ευρωπαϊκό Κοινωνικό Ταμείο



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ΕΚΠΑΙΔΕΥΣΗ ΚΑΙ ΔΙΑ ΒΙΟΥ ΜΑΘΗΣΗ
επένδυση στην υποτροφία της γηώσης

ΥΠΟΥΡΓΕΙΟ ΠΑΙΔΕΙΑΣ ΚΑΙ ΘΡΗΣΚΕΥΜΑΤΩΝ
ΕΙΔΙΚΗ ΥΠΗΡΕΣΙΑ ΔΙΑΧΕΙΡΙΣΗΣ

Με τη συγχρηματοδότηση της Ελλάδας και της Ευρωπαϊκής Ένωσης



ΕΣΠΑ
2007-2013
πρόγραμμα για την ανάπτυξη
ΕΥΡΩΠΑΪΚΟ ΚΟΙΝΩΝΙΚΟ ΤΑΜΕΙΟ